

High yielding  
conversion of  
xylose to furfural  
via boronate  
esters

Peter van der Wal  
University of Twente  
p.j.vanderwal@utwente.nl



8  
Sam Houston  
Tollway  
EXIT 2 MILES

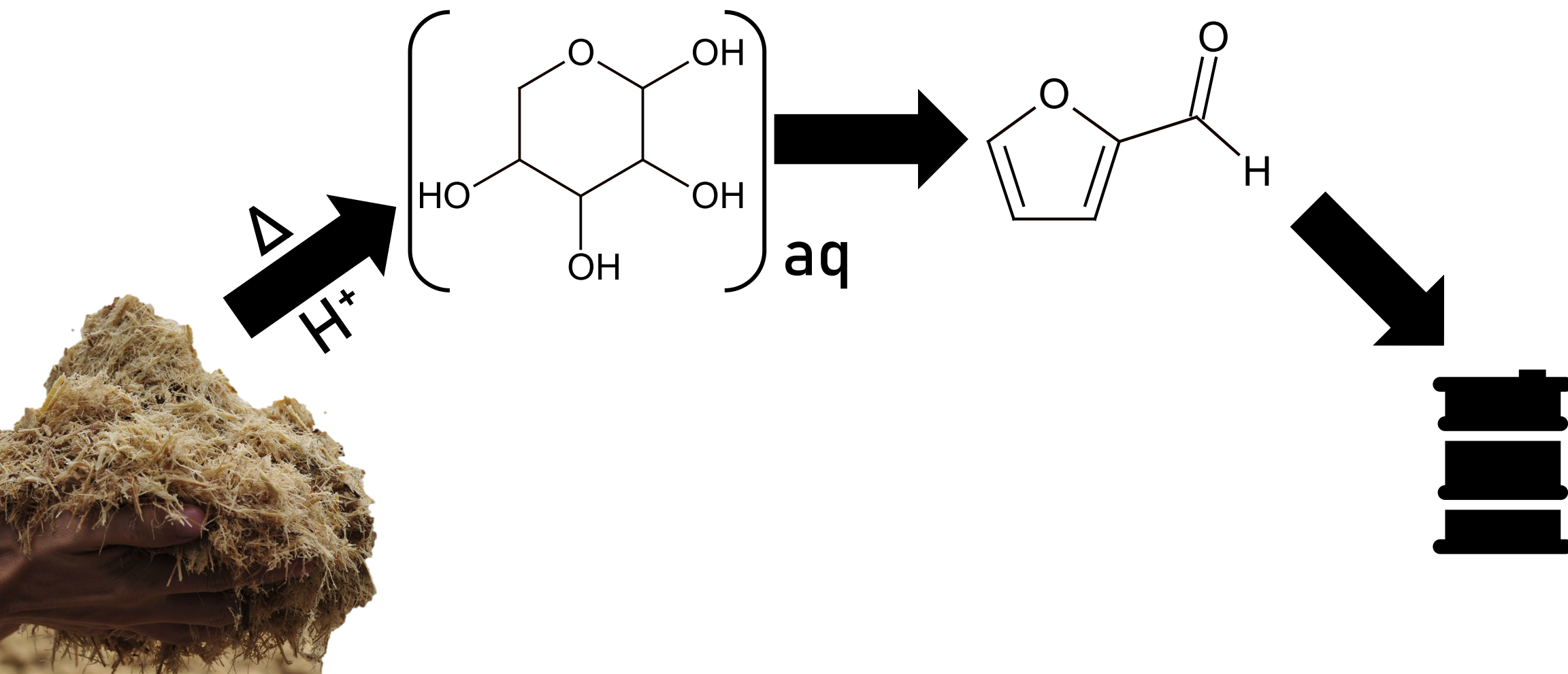
[Blacked out sign]



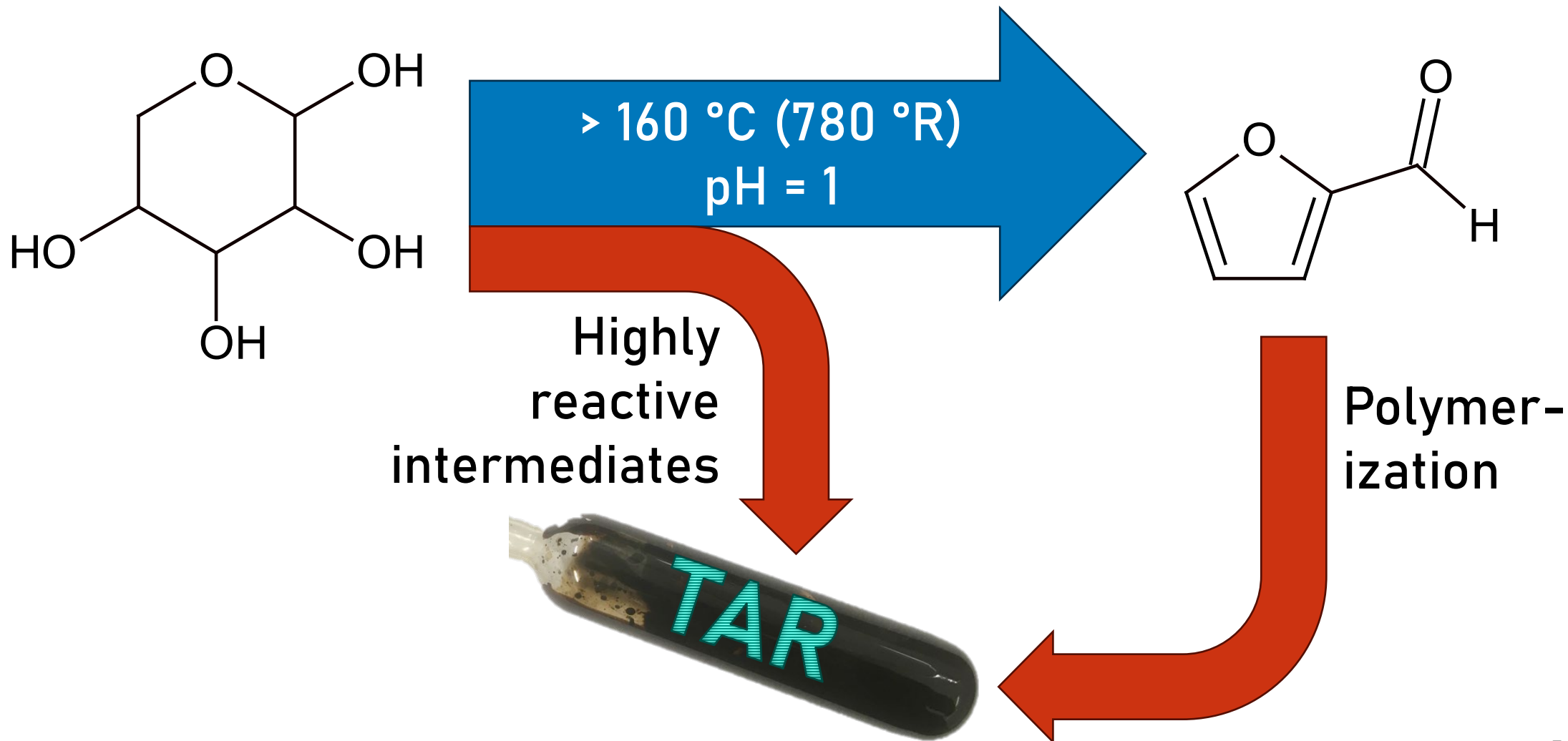
NO STOPPING  
OR PARKING



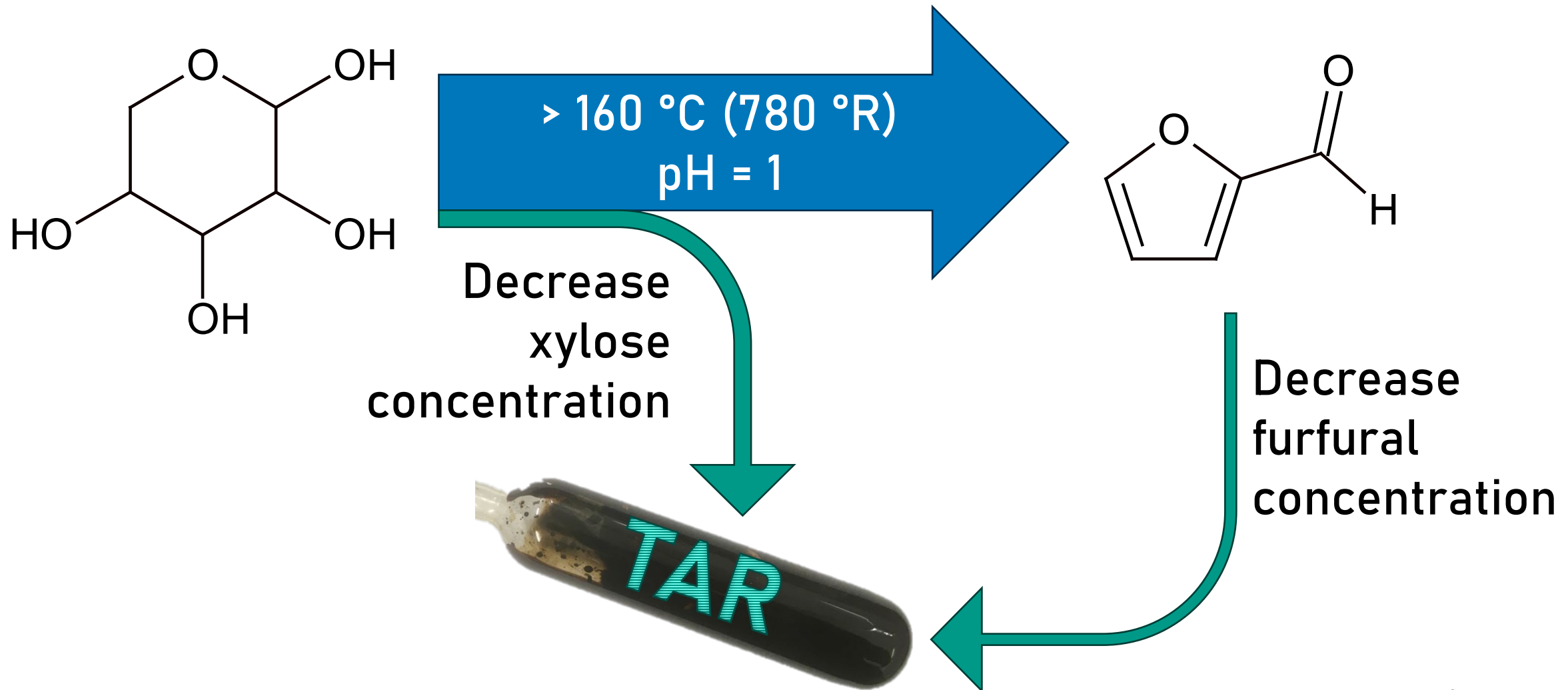
# From by-product to platform chemical



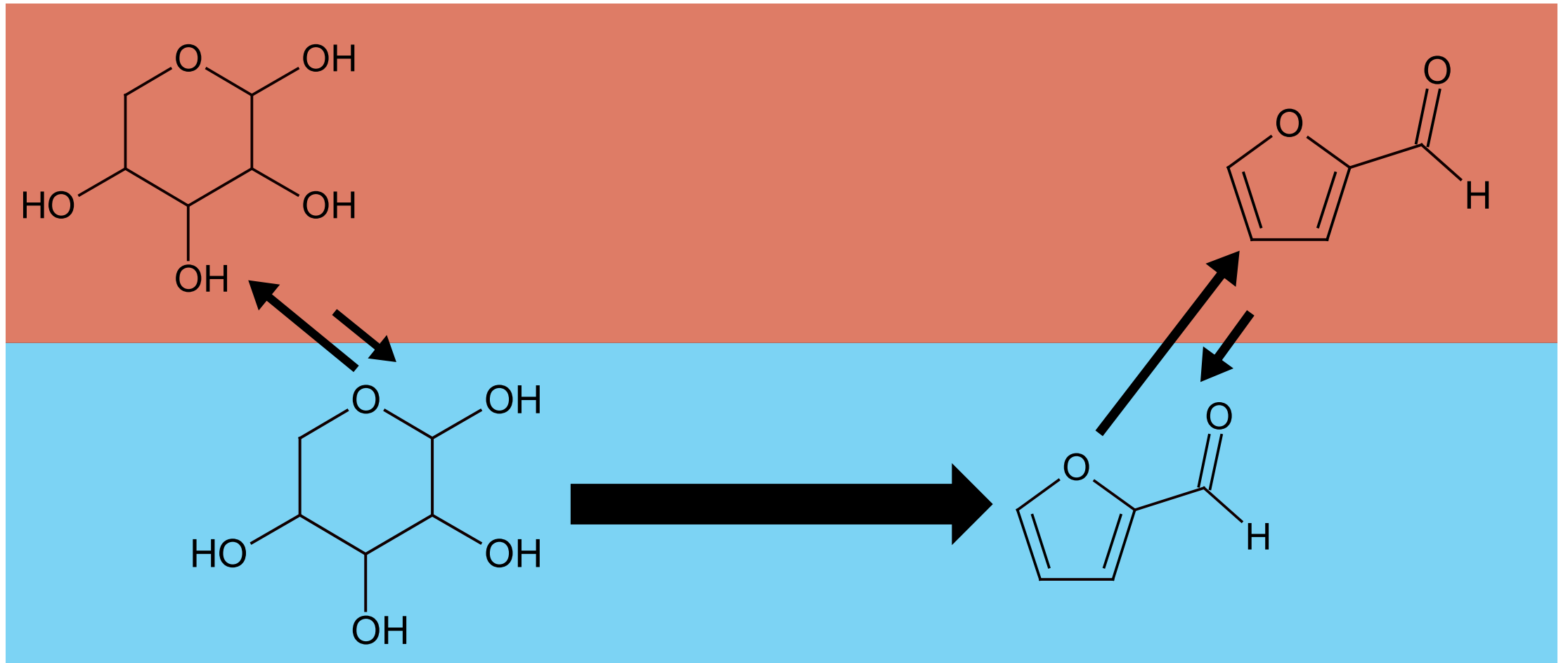
# Current practice



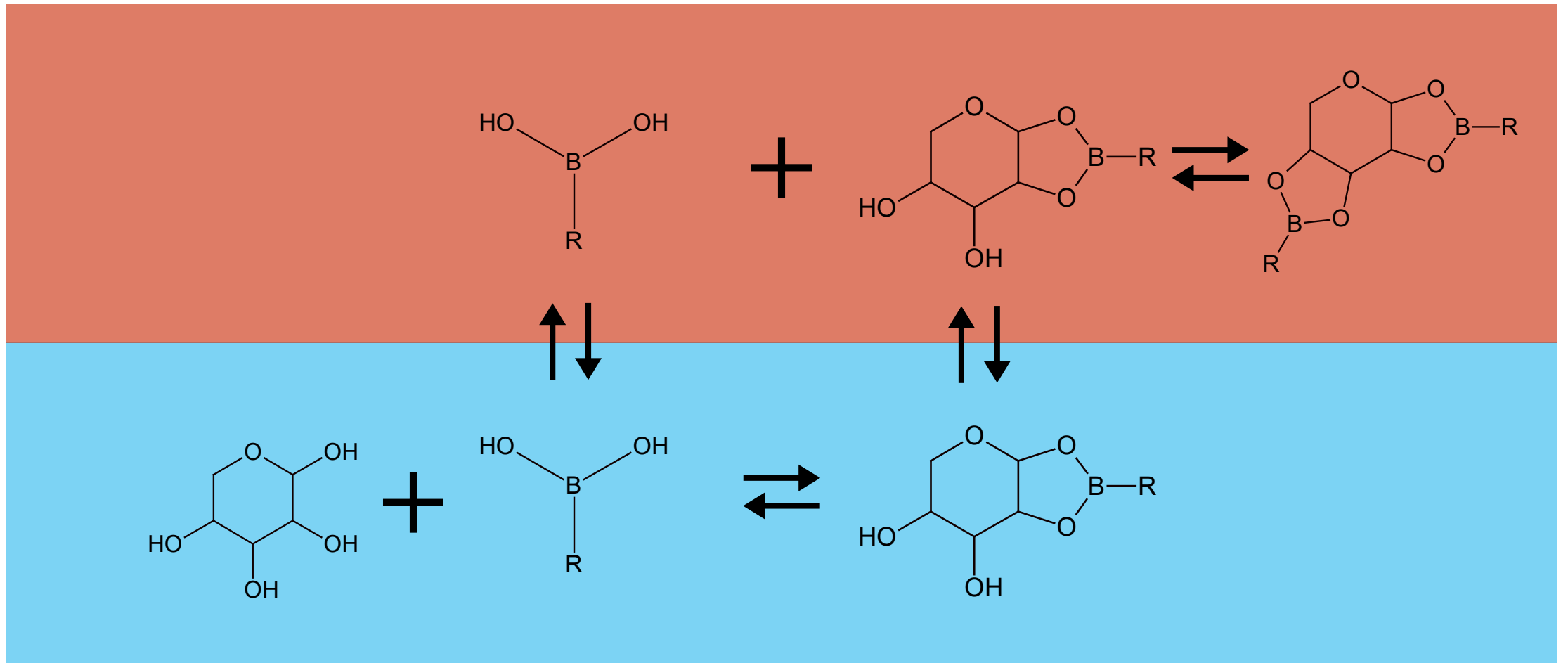
# Alternatively



# Realizing low concentrations

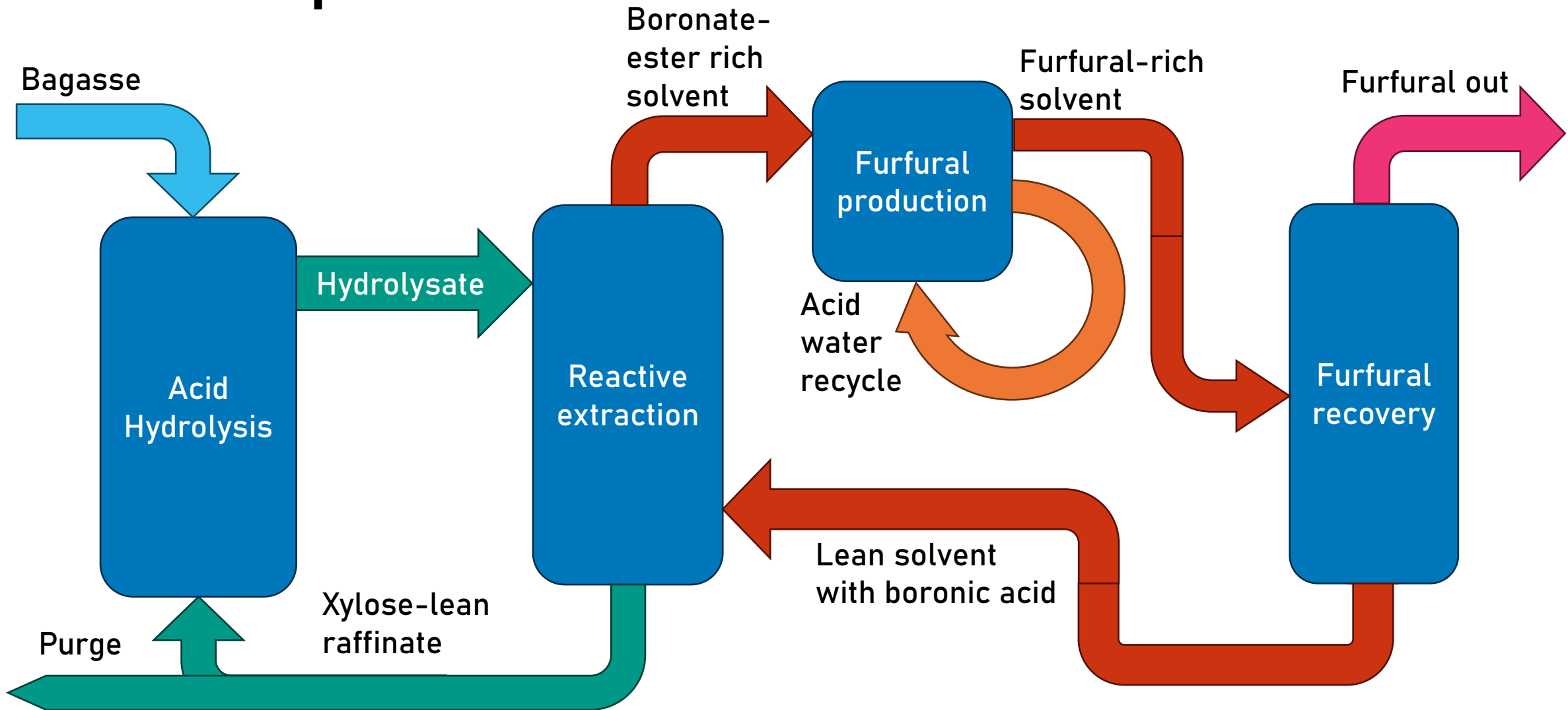


# “Dissolving water into oil”





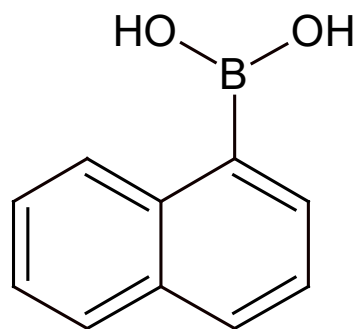
# Overall process



# Choice of chemicals

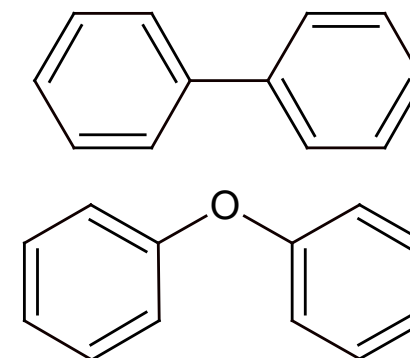
## Boronic acid

- Apolar side group
  - High partition of ester to organic phase
  - Low solubility in water
- High reaction rate in esterification
- Aromatic



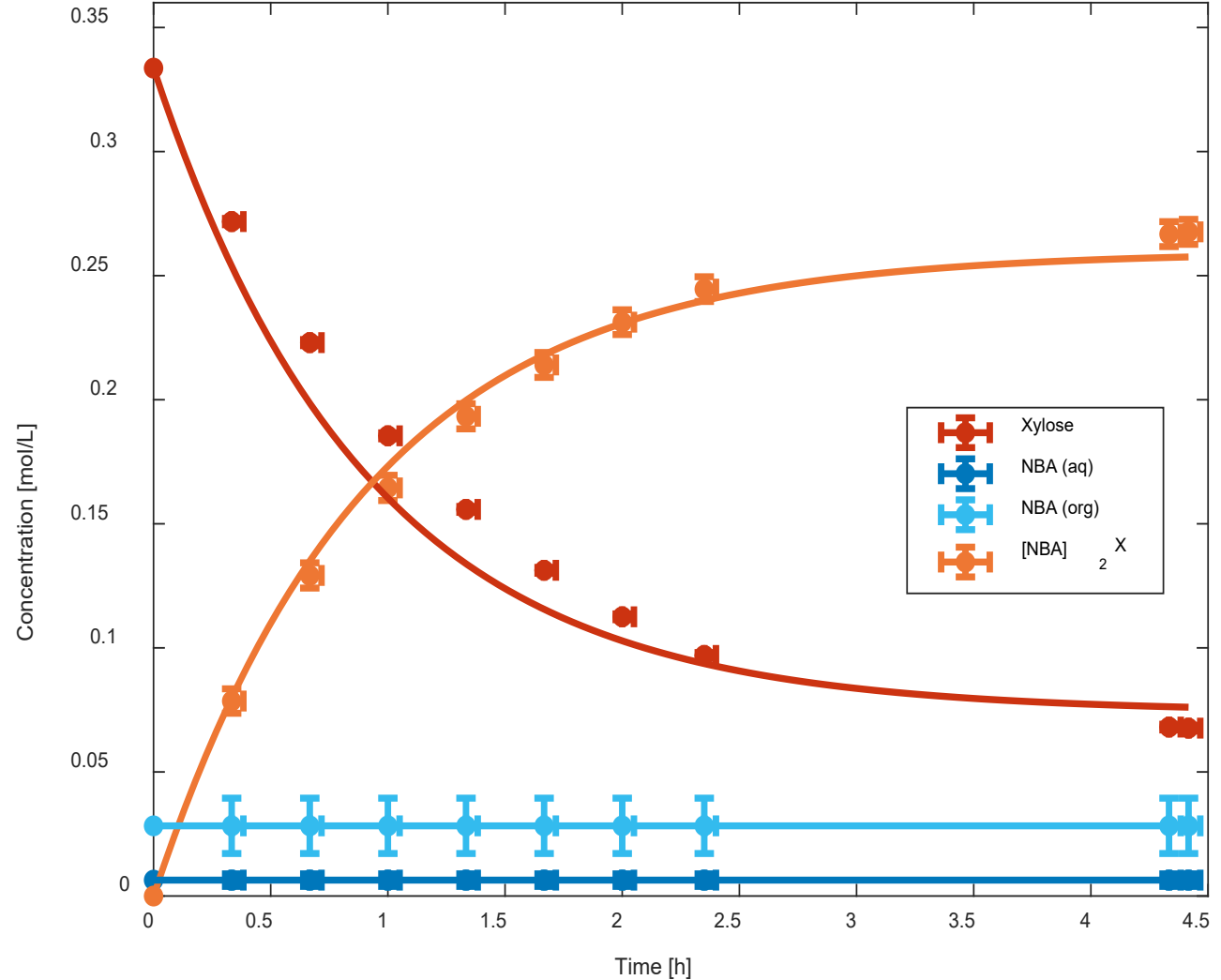
## Solvent

- Aromatic
- Boiling point above furfural (162 °C)



# Extraction experiments

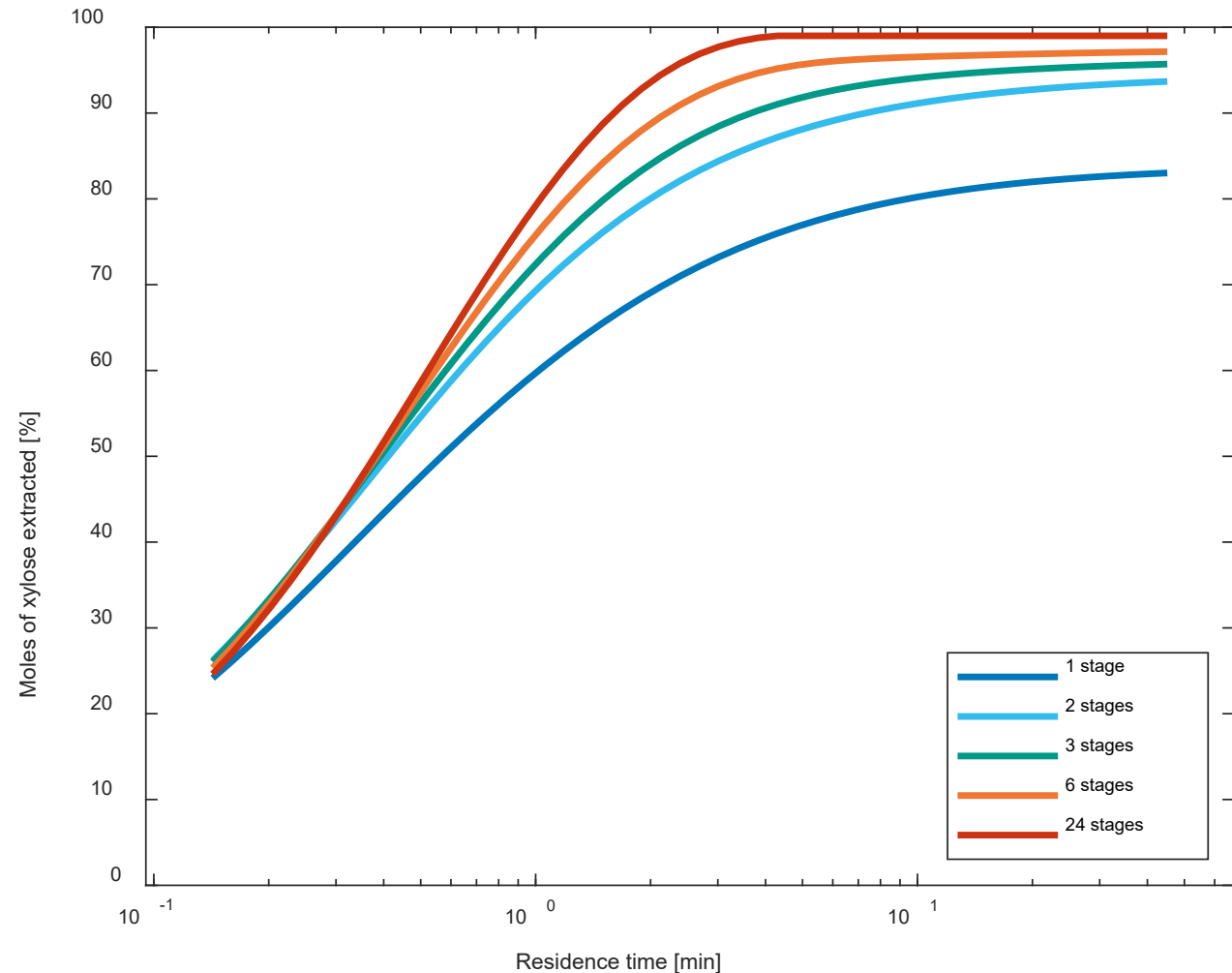
- Measured kinetics and equilibria from 20 to 90 °C
- Higher T favors kinetics and equilibrium
  - >80 % extracted in a few minutes at 90 °C



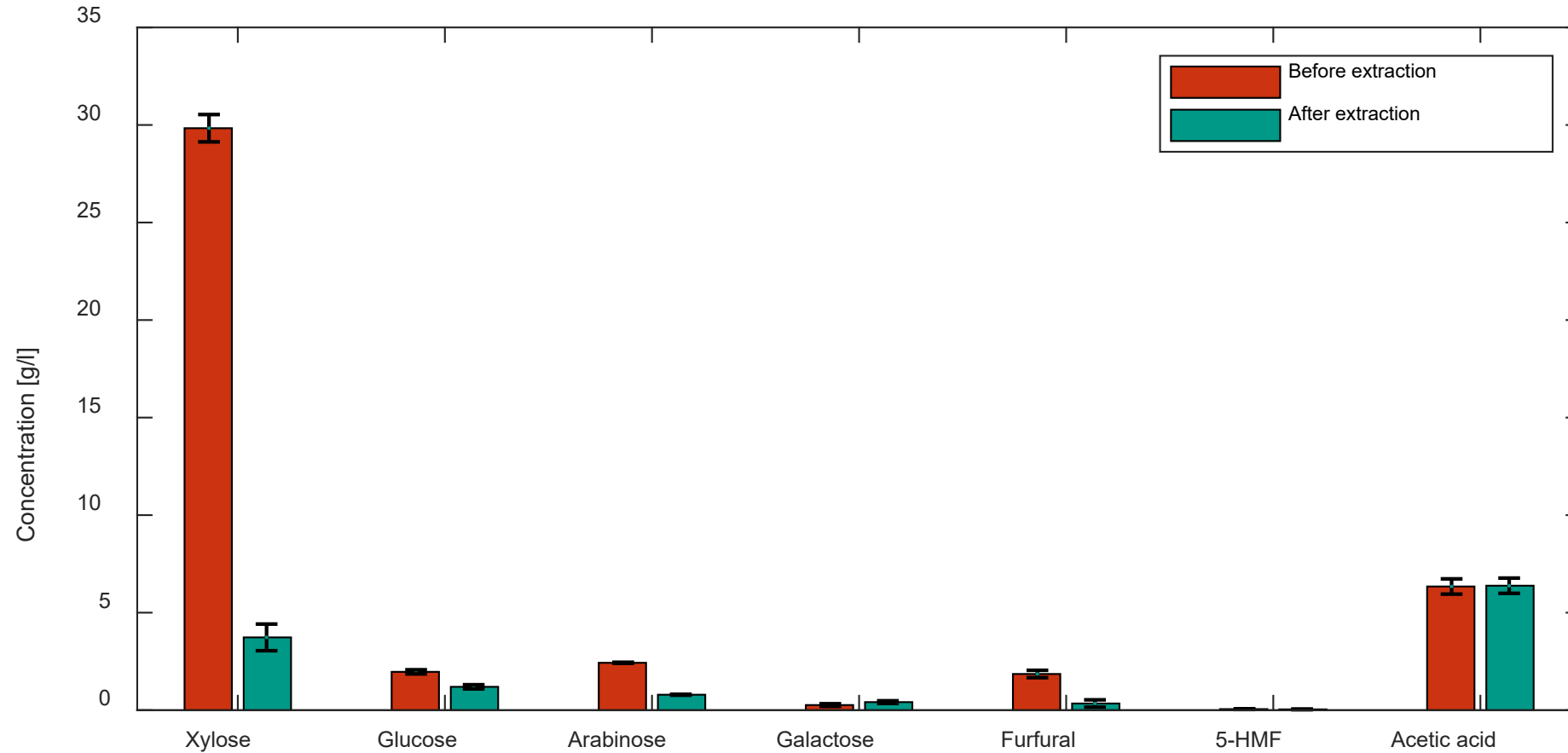
# Extraction model

---

- CSTR model using measured kinetics and equilibria
- Designed a mixer-settler cascade for 90 % xylose recovery
  - Need three stages and 4 minutes
- Need to crystallize NBA out from raffinate

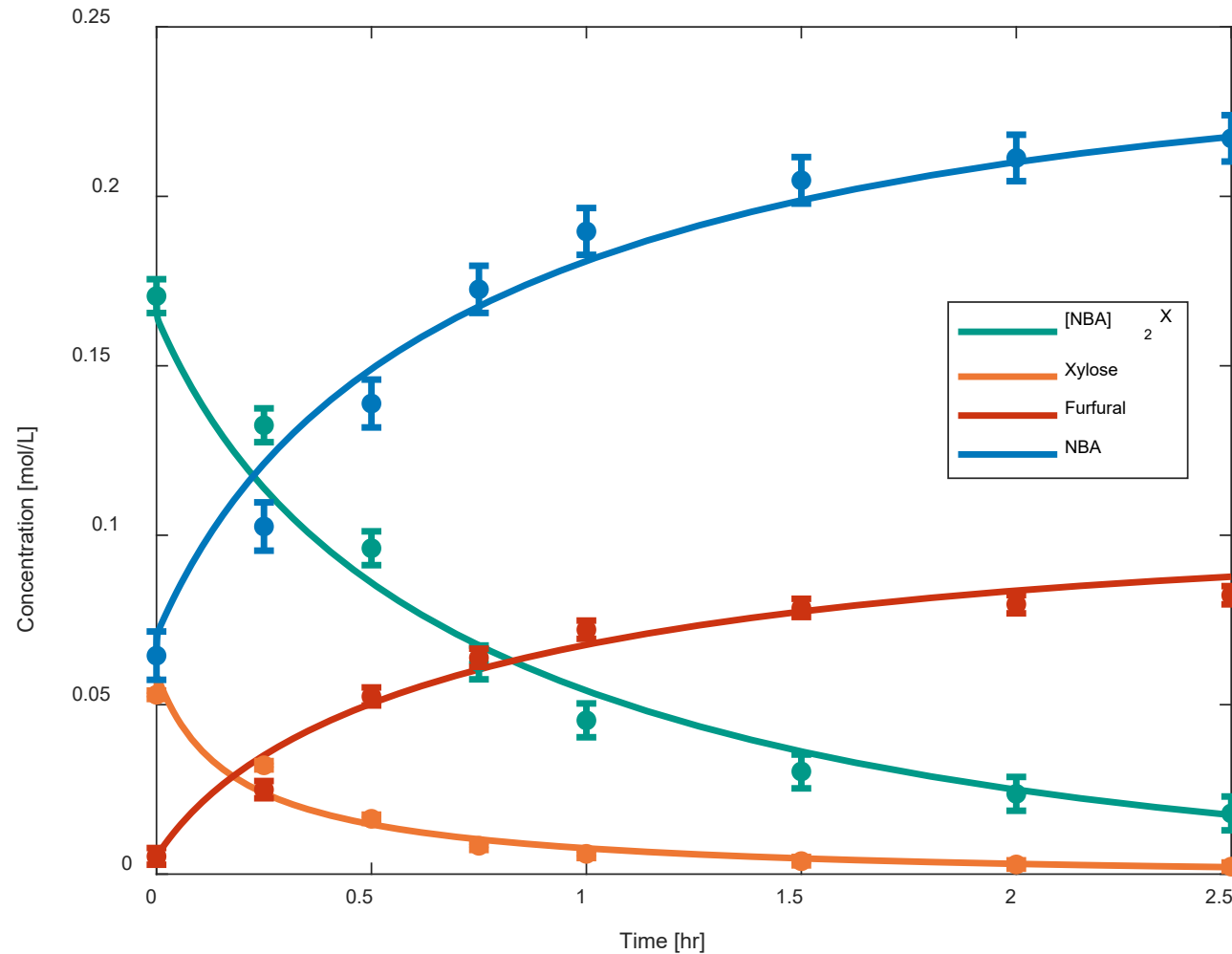


# What about real (TC)biomass?



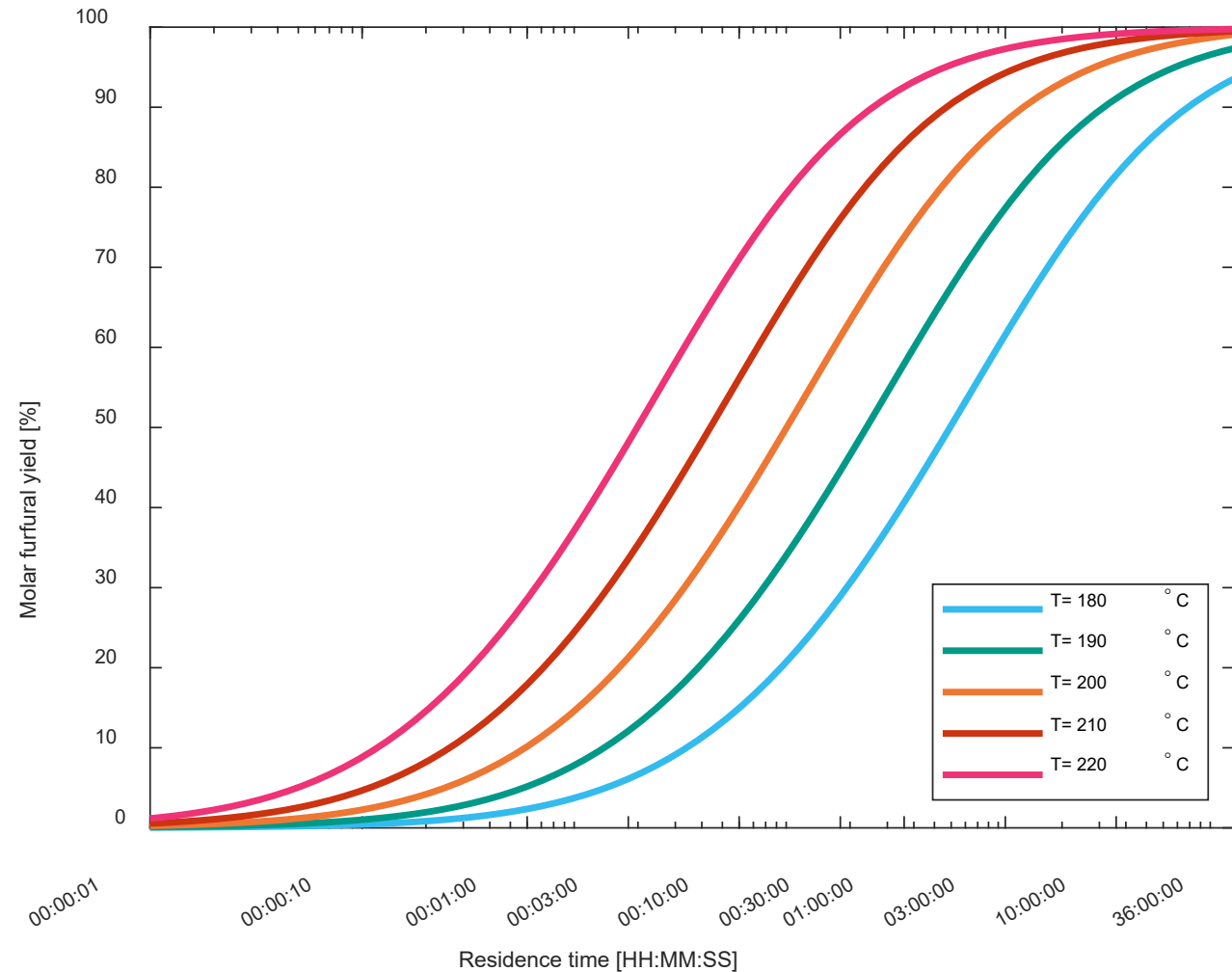
# Furfural production experiments

- Successful production of furfural
- Also proven for hydrolysate
- >70 % molar selectivity batch-wise
- Selectivity varies little with temperature



# Furfural production model

- Modelled a CSTR
- >90 % selectivity possible
- Construction is ongoing for experimental validation




# Conclusion

---

- State-of-art yield of furfural is low
- Boronate ester improvement
- Proven reactive extraction
- Promising conversion







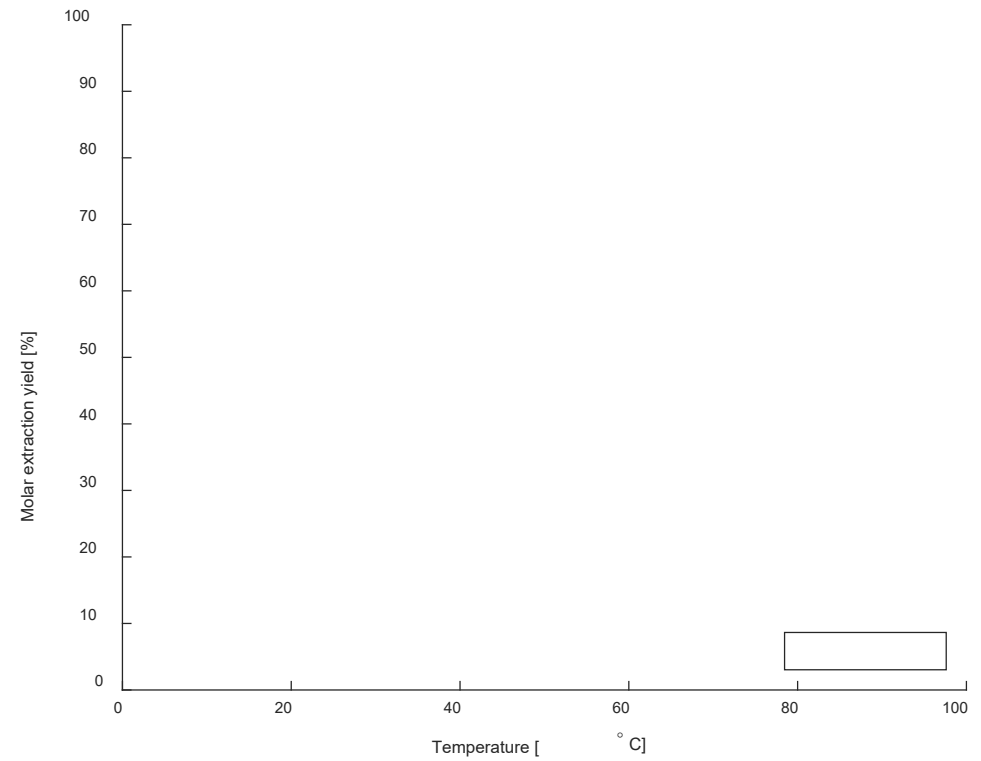
# High yielding conversion of xylose to furfural via boronate esters

Peter van der Wal  
University of Twente  
p.j.vanderwal@utwente.nl

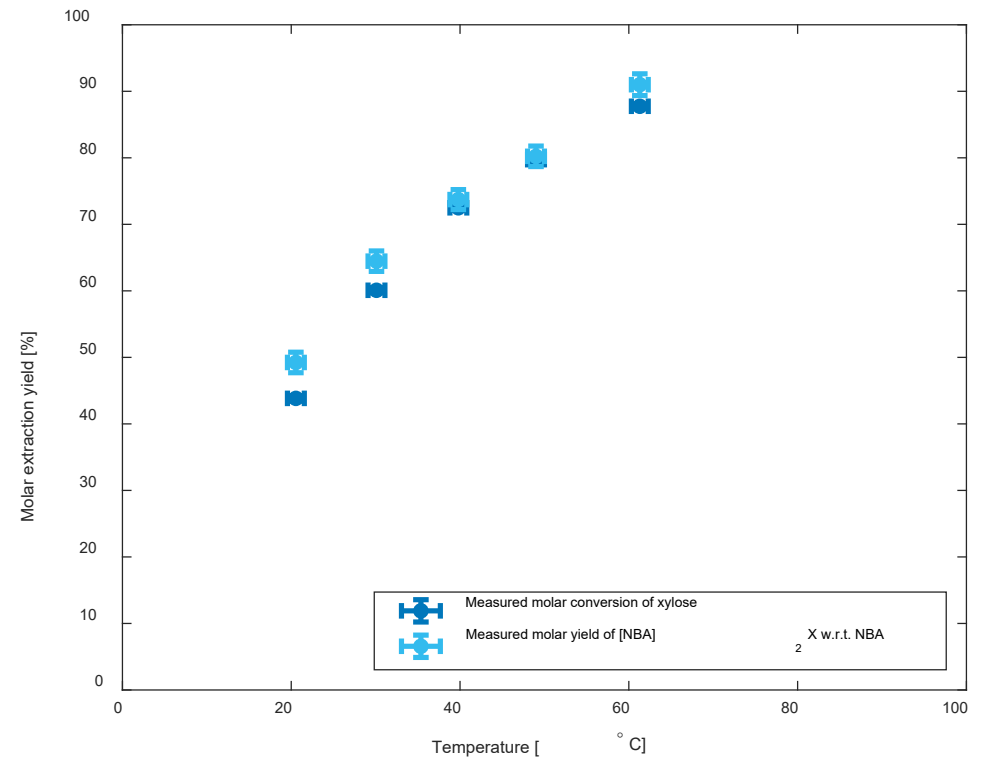
# What about real (TC)biomass?

	Initial amount (g/L)	Extracted per pass (wt.%)
Xylose	29.83±0.70	87.5±2.3
Glucose	1.97±0.11	39.0±6.4
Arabinose	2.43±0.03	67.4±1.1
Galactose	0.26±0.07	No significant change
Furfural	1.87±0.19	81.5±10.3
5-HMF	0.06±0.02	No significant change
Acetic acid	6.34±0.39	No significant change

# A closer look at the equilibrium

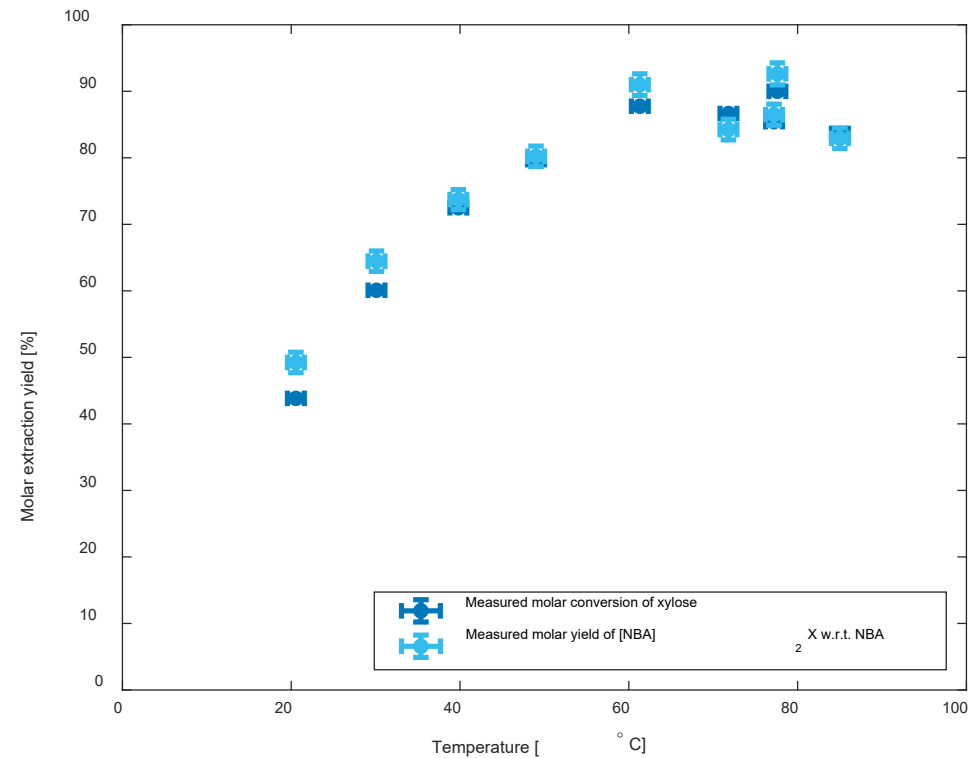


# A closer look at the equilibrium



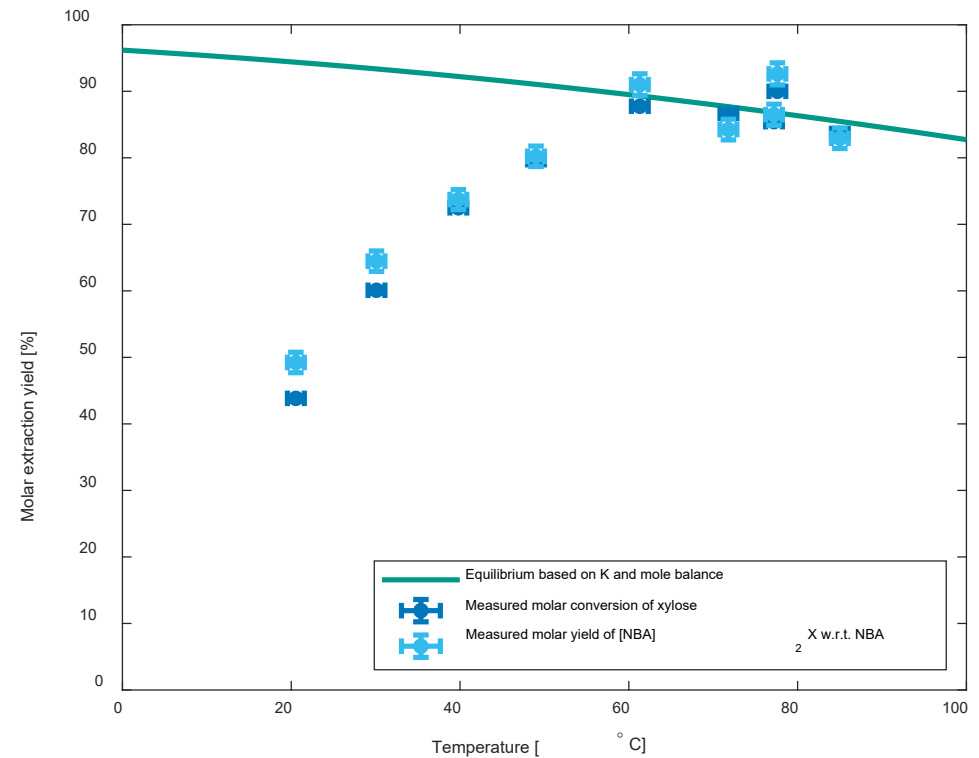
# A closer look at the equilibrium

$$\bullet K = \frac{c_{[NBA]_2X}^{org}}{c_X^{aq} c_{NBA}^{aq} c_{NBA}^{org}}$$

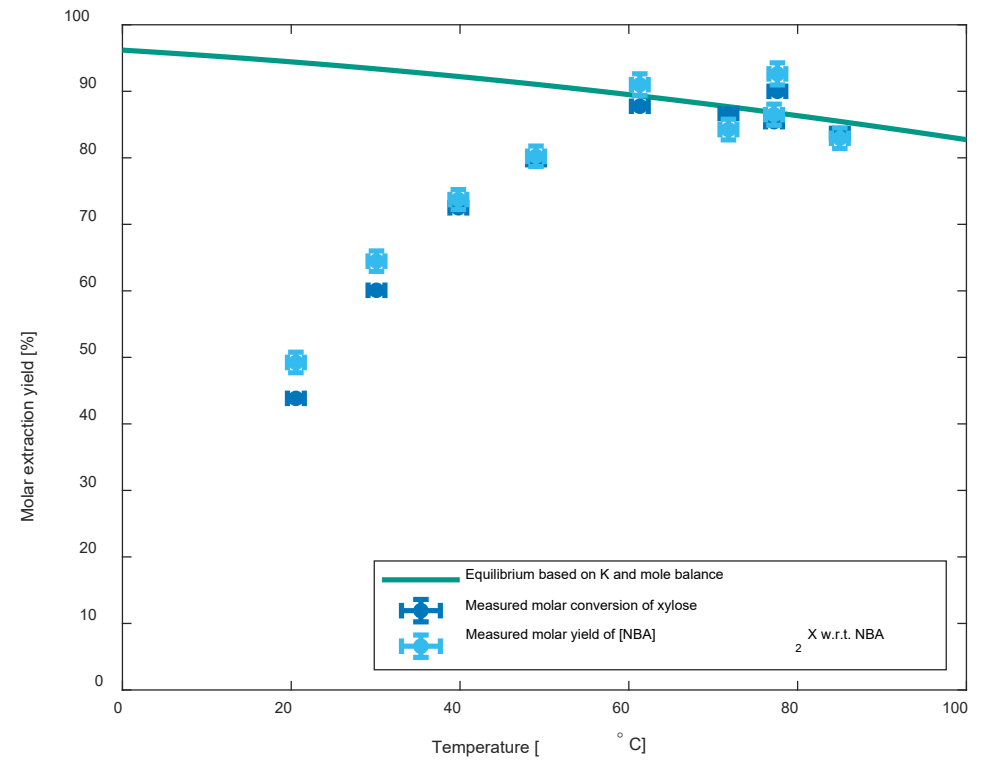
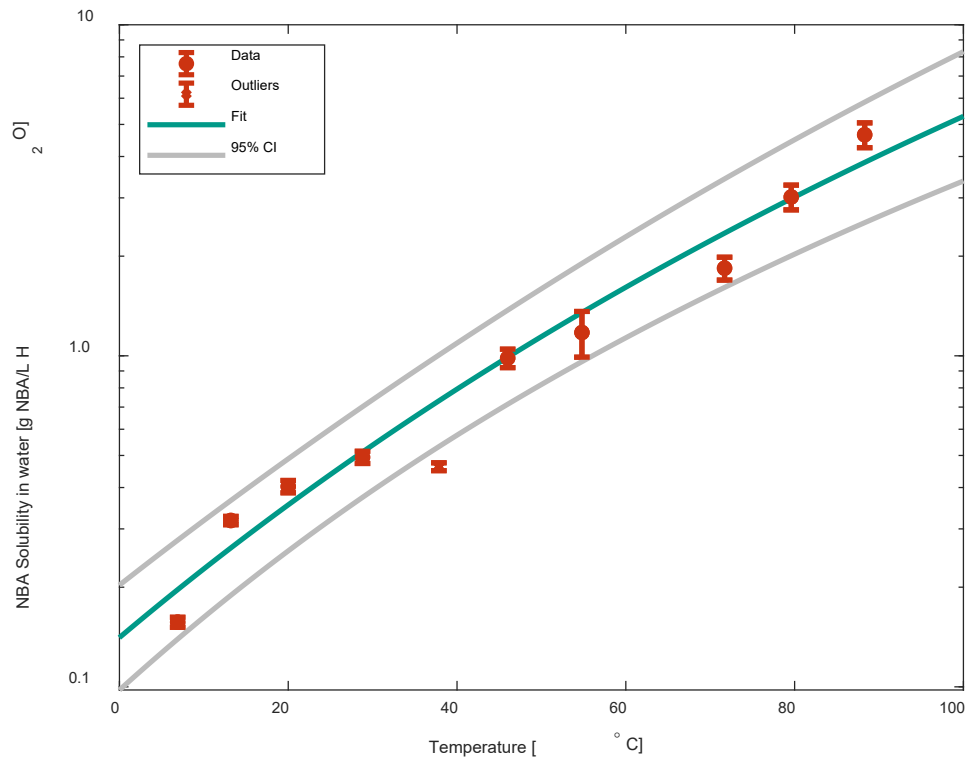


# A closer look at the equilibrium

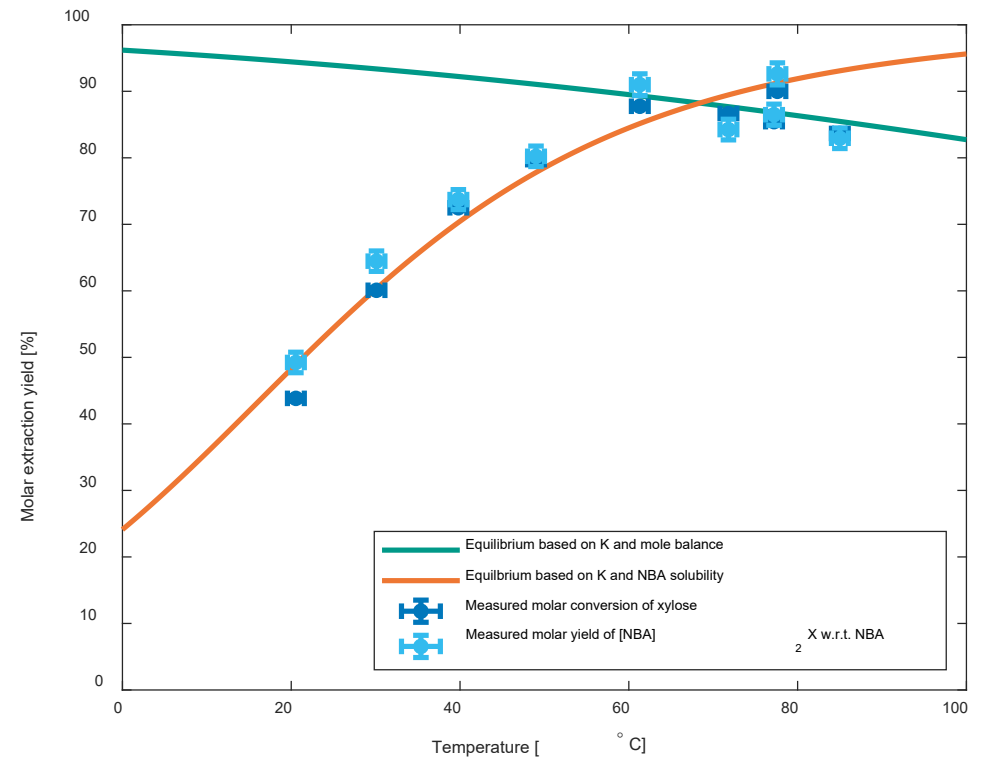
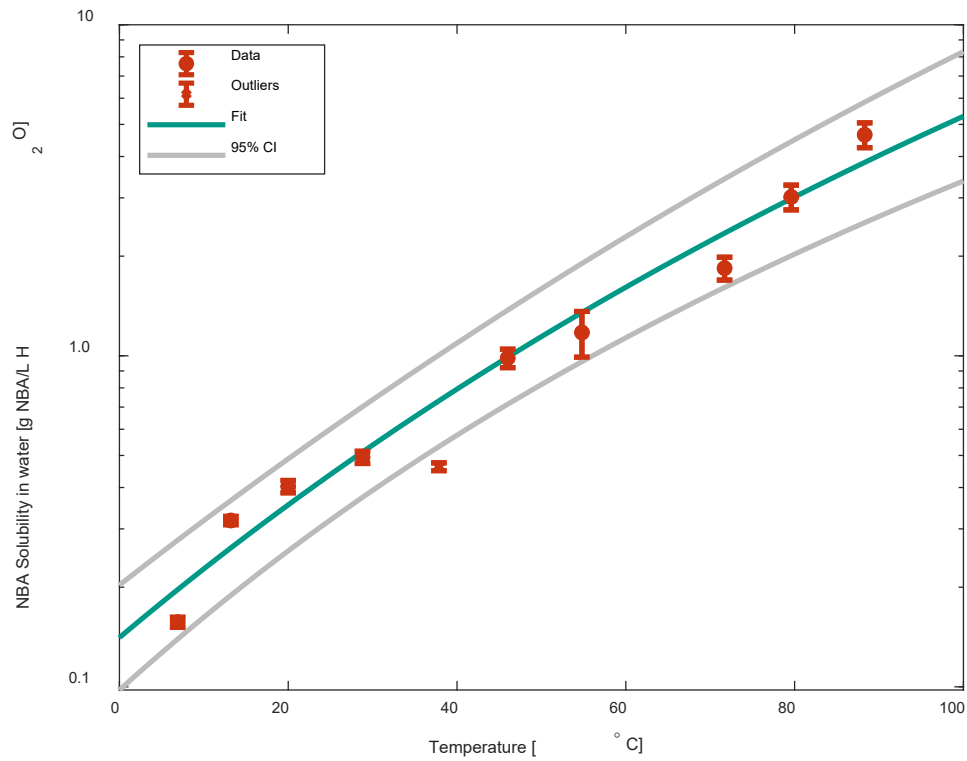
$$\bullet K = \frac{c_{[NBA]_2X}^{org}}{c_X^{aq} c_{NBA}^{aq} c_{NBA}^{org}}$$



# A closer look at the equilibrium

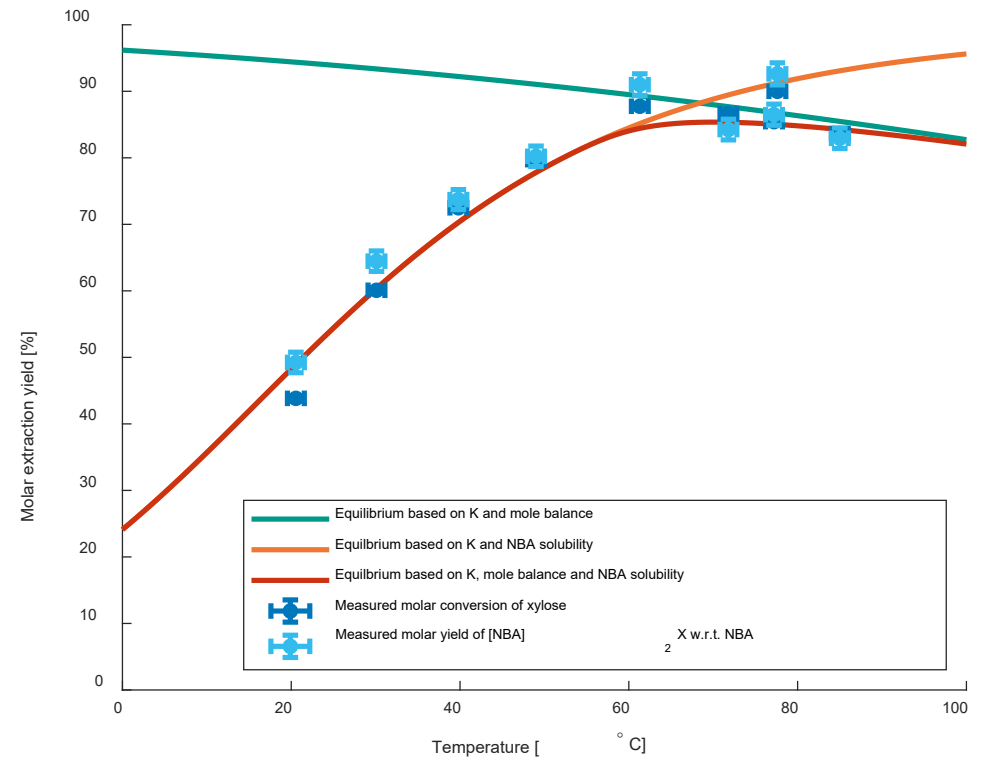
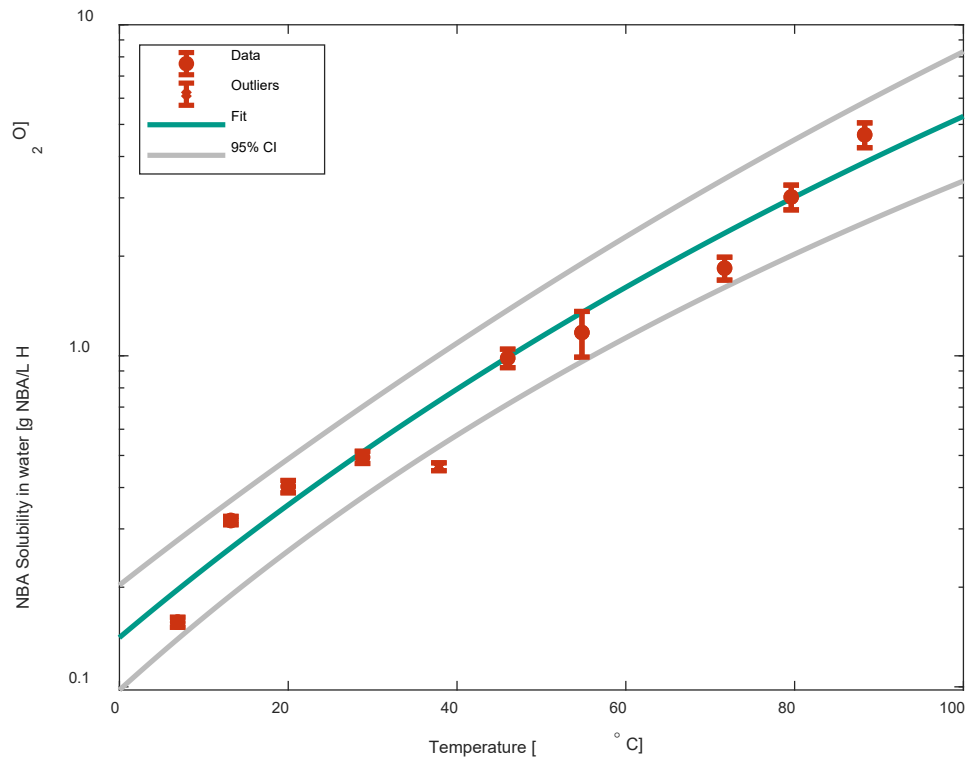


# A closer look at the equilibrium

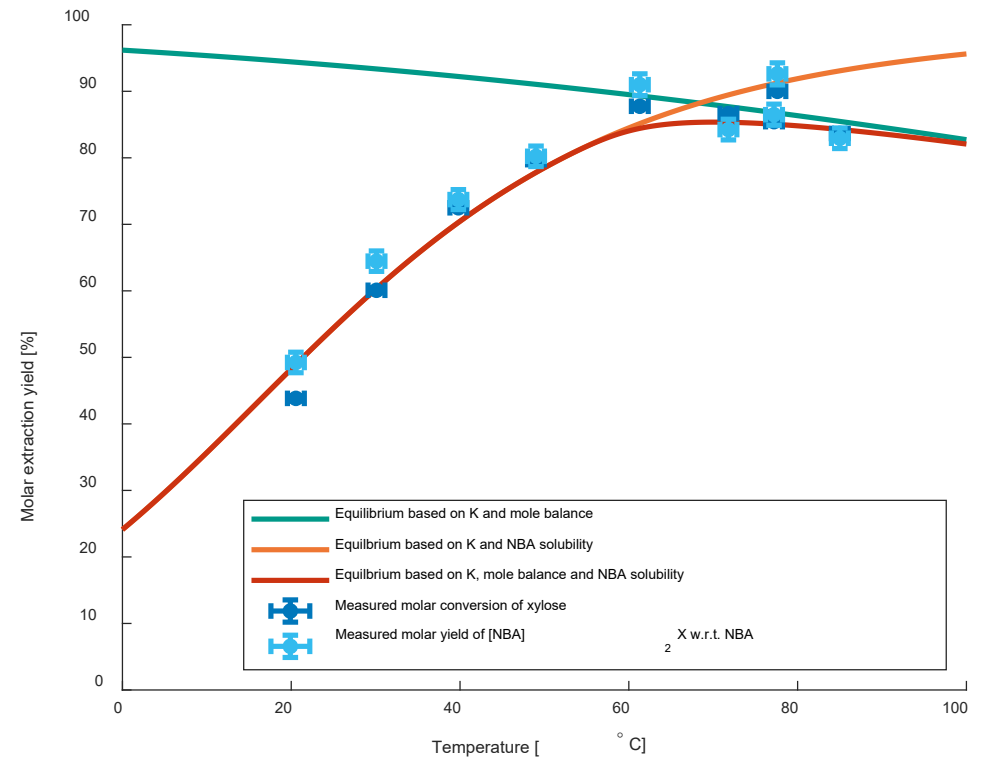
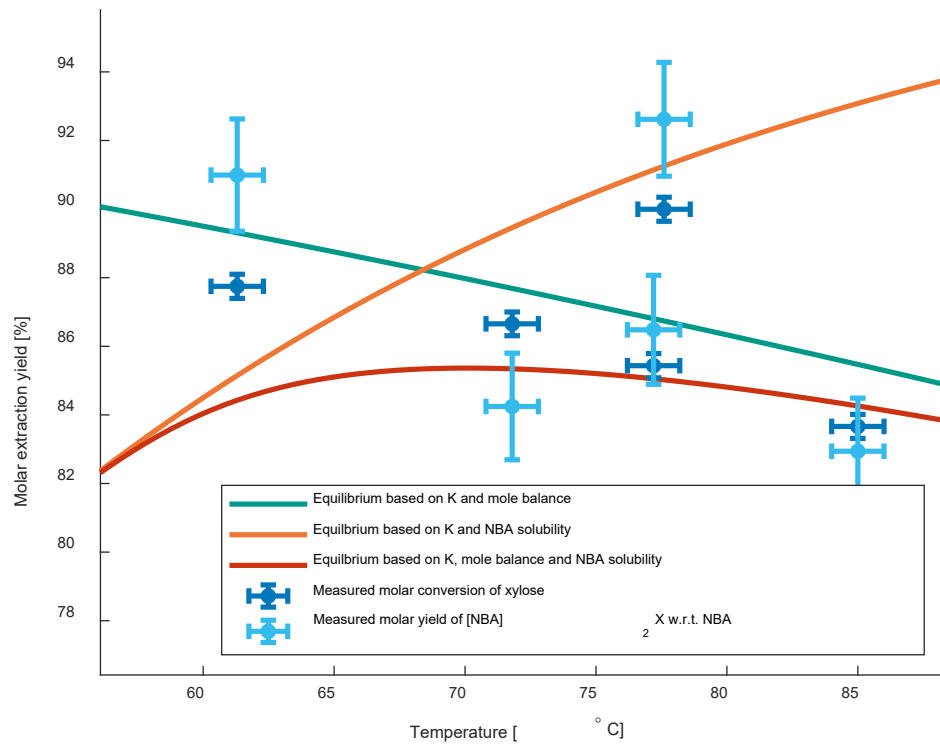





# A closer look at the equilibrium

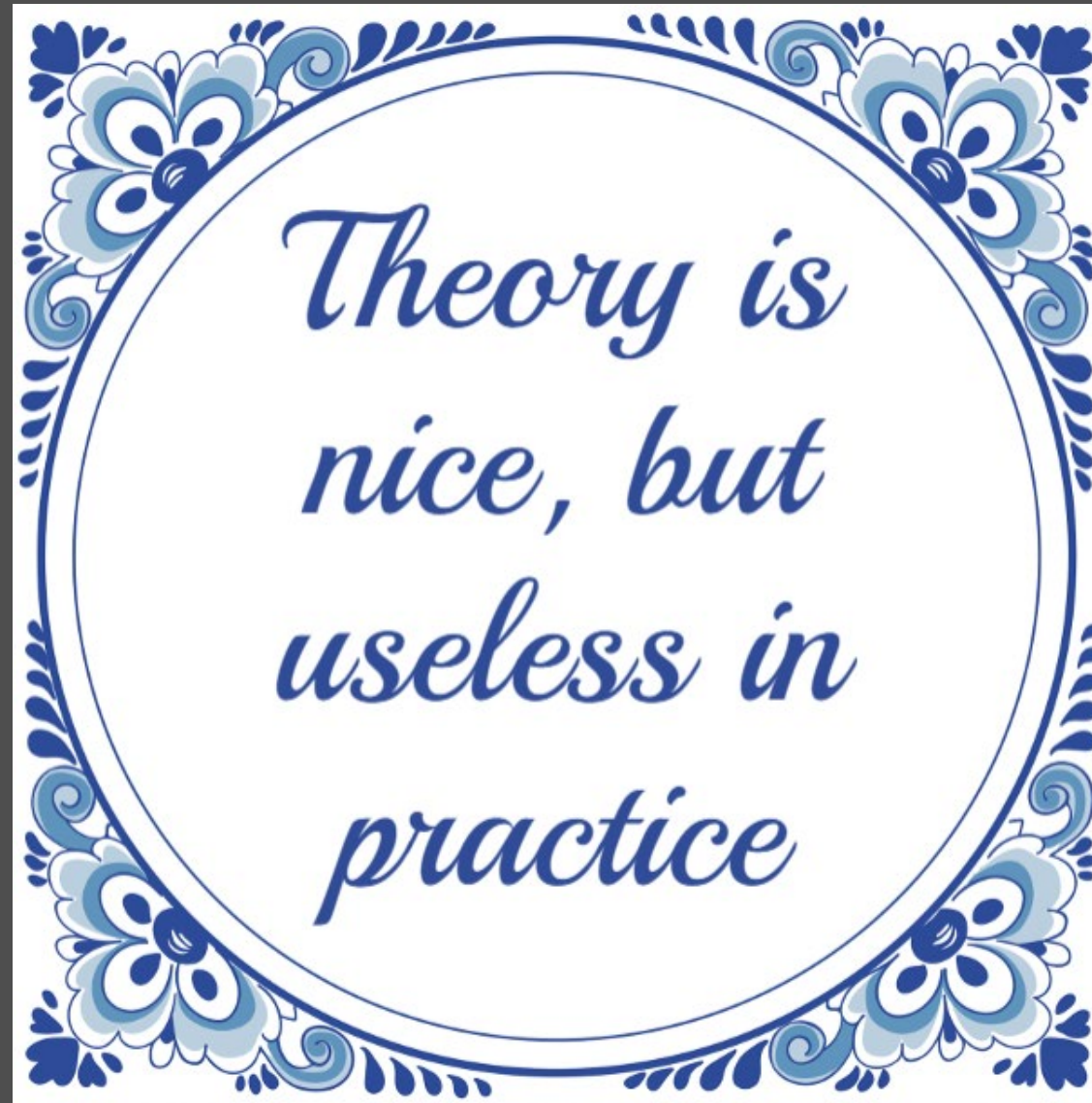


# A closer look at the equilibrium



A decorative border in shades of blue, featuring stylized flowers and scrolls, framing the text.

*Theorie is mooi,  
maar in de praktijk  
heb je er niets aan*

A decorative border in shades of blue, featuring stylized flowers and scrolls, framing the text.

*Theory is  
nice, but  
useless in  
practice*