### **Catalyst For The Reformation Of Biomass Pyrolysis Gas And Their Effect On Methanation**



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tcbiomass2024

## A DCL Technology Group Company





# First emissions controls developed for Underground Mining in 1986.









# **40 Years of Energy Efficiency** Trusted by power producers to ensure cost effective, reliable emission reduction solutions for a broad range of industries.





Mobile & Stationary Emissions

Cogeneration Energy Efficiency



Biogas Conditioning & RNG Upgrading



## Expertise in emissions controls, nanomaterials, catalyst & energy efficiency.

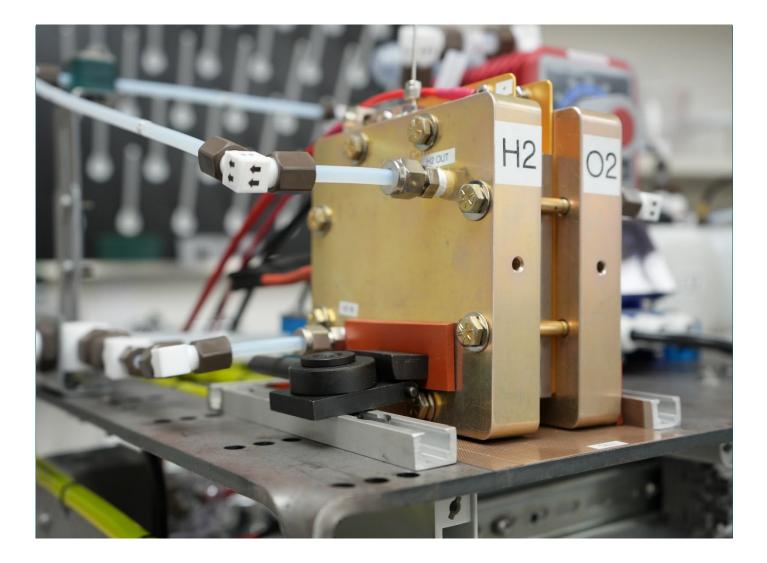


40 Years of R&D, Engineering, & Manufacturing.

# **Decarbonization Solutions**

DCL's expertise in designing & manufacturing high-quality components, for maximum efficiency & performance for clean energy applications, including Hydrogen CHP systems.





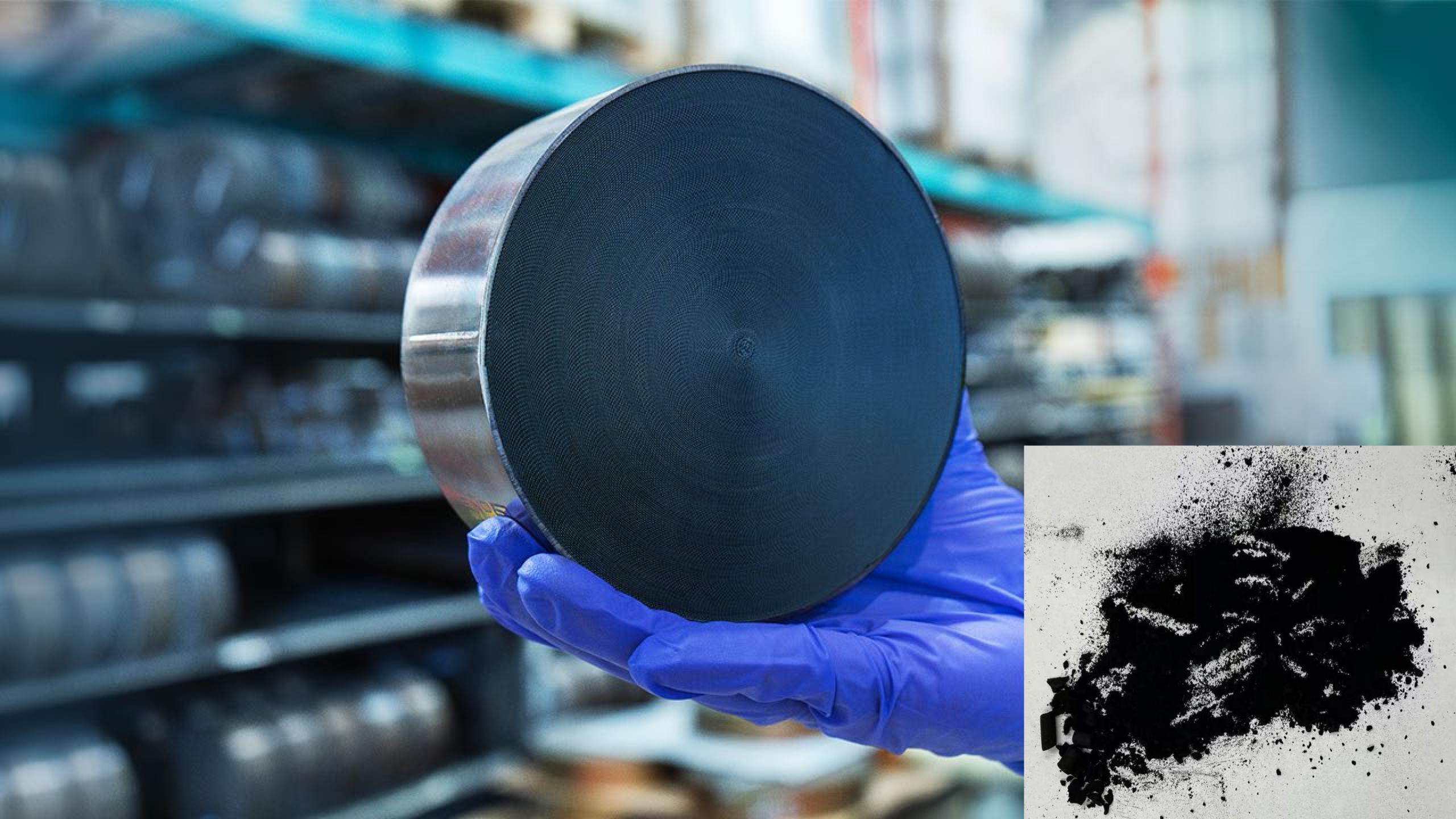
**Fuel Cells** 

Electrolyzers

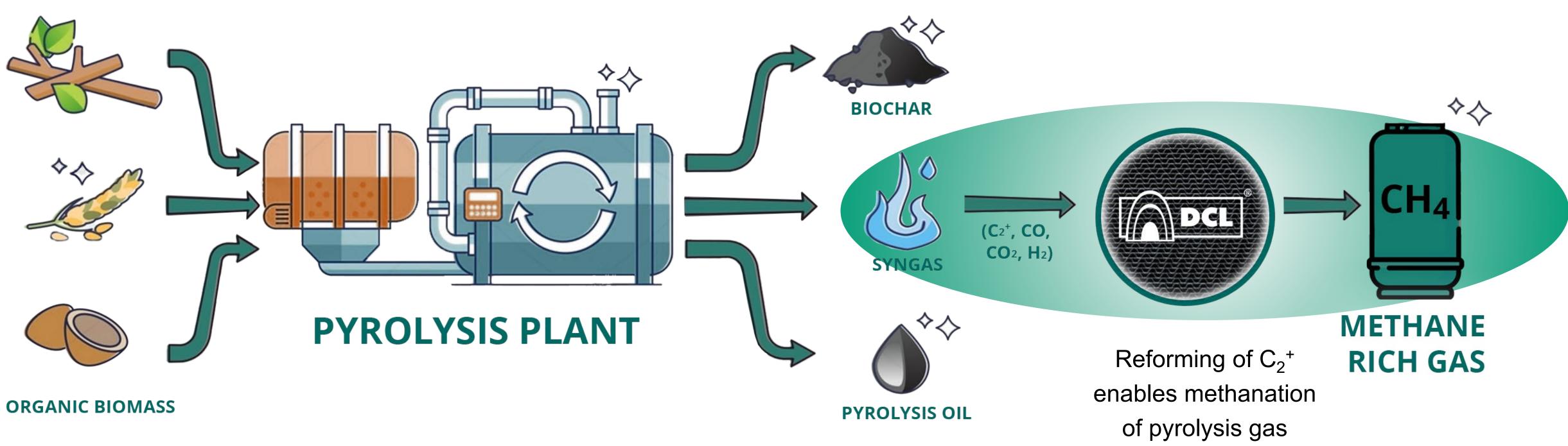




Reforming & Methanation Catalysts



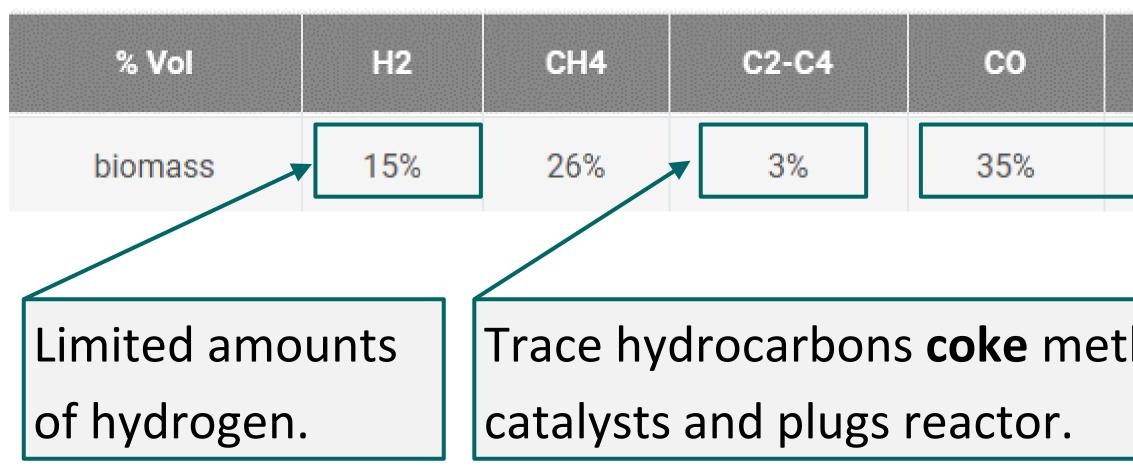
### **Pyrolysis Gas – Methanation**







### **Composition & Challenges of Pyrolysis** Gas



### $CO/CO_2$ reacts with $H_2$ to produce CH<sub>4</sub>

\*LHV of Methane is 2-3X greater than biomass

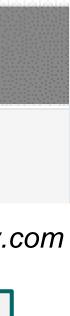




C02	N2	LHV [MJ/Nm3]	kWh/tonne inlet
17%	4%	17.10	2600
		Source	ce: https://www.biogreen-energy.o
thanatio			
	r	nethanation or WGS to recycle carbon.	

Typical Ni Based Catalyst for Methanation:

- Susceptible to Sulfur poisoning  $\bullet$
- Significant Coke accumulation due to HCs lacksquare
- Max operating temperature 500° 550°C  $\bullet$



### Pyrolysis Gas Pre-Treatment

	And	
<b>General Composition</b>		
H <sub>2</sub>	~15- 40 %	
CO <sub>2</sub>	~10 - 20 %	
CO	~20 - 35 %	
$CH_4$	~10 - 30%	
$C_2H_4$	~3 %	
$C_2H_6$	~0.3 %	
$C_3H_6$	~0.4 %	
$C_3H_8$	~0.4 %	
TARS	~0.2 %	
DUST	~0.2 %	
NH <sub>3</sub>	~0.3 %	
H <sub>2</sub> S	~0.1 %	

#### Temperature >800°C



**DCL Technology Group's gas** pre-treatment equipment in operation on W2E and other biogas applications



### **Treatment Processes For Pyrolysis Gas**

**Pre-Treatment** eliminates TARS, DUST, NH<sub>3</sub>, H<sub>2</sub>S

- Limited by hydrogen content in pyrolysis gas.
- Straightforward process if green hydrogen is available.

### **Pre-Treatment** eliminates TARS, DUST, NH<sub>3</sub>, H<sub>2</sub>S

Water Gas Shift (WGS) converts CO to CO<sub>2</sub> + generates H<sub>2</sub>

- Increase H<sub>2</sub> concentration using the WGS reaction.
- Requires energy-intensive process with careful energy management.
- Green hydrogen availability boosts CH<sub>4</sub> yield.

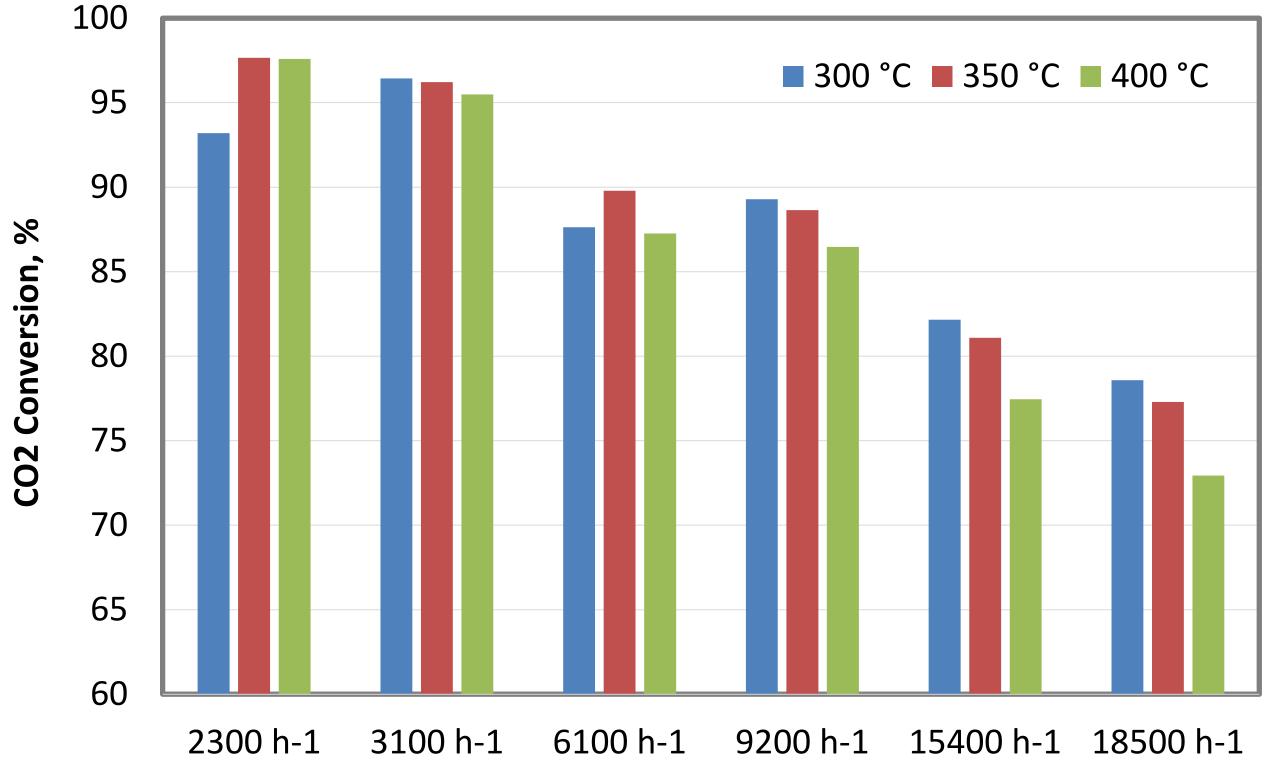


**Methanation** converts CO and CO<sub>2</sub> into CH<sub>4</sub>

> **Methanation** converts CO and CO<sub>2</sub> into CH<sub>4</sub>



### Baseline: CO<sub>2</sub> Methanation



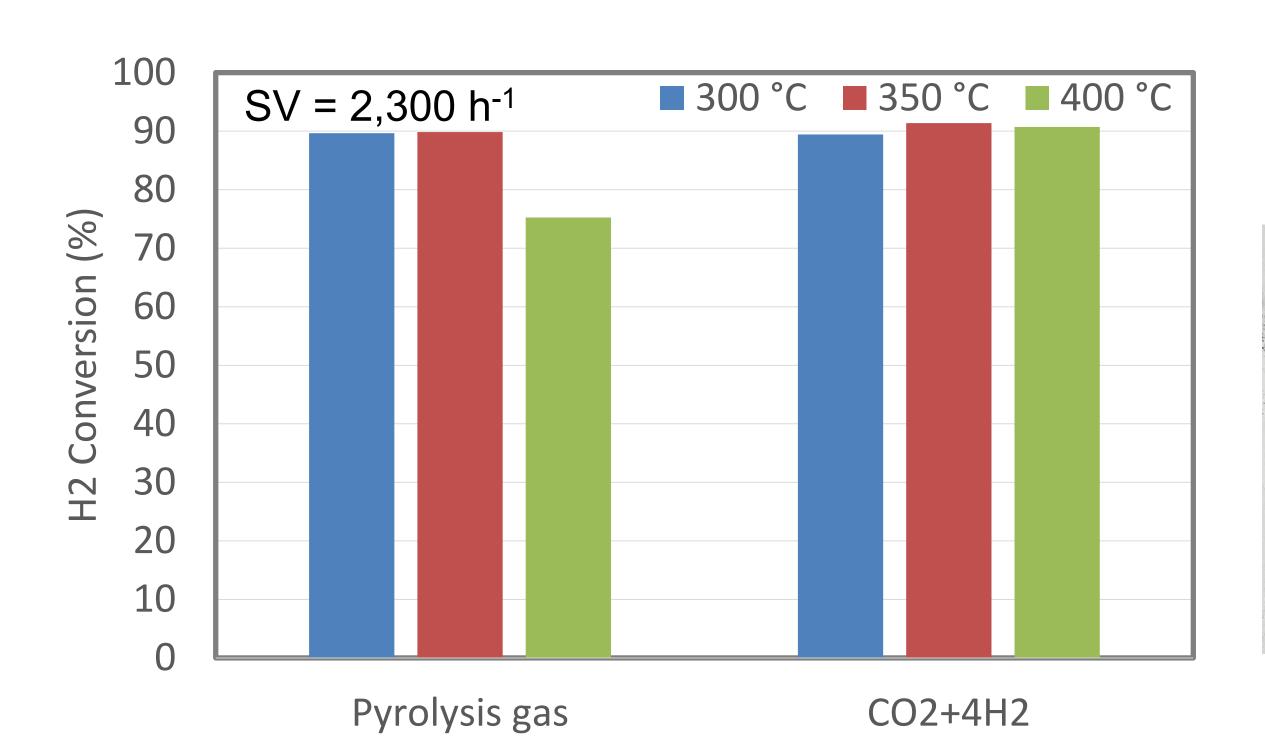


Equilibrium conversion is achieved at 300 - 350°C with a Ni catalyst at 2,000 - 3,000 h<sup>-1</sup> space velocity.

> Ni Based Catalyst for Methanation  $CO_2 + 4H_2 = CH_4 + 2H_2O$  $CO_2: H_2 = 1:4$



### Pyrolysis gas Methanation on Ni based catalyst



**Pre-Treatment** 

#### \*Coke formations block **Reactor within 4hrs**



Gas Comp.	Vol, %
H <sub>2</sub>	19
CO	35
CO <sub>2</sub>	21
$CH_4$	20
$C_2H_6$	1.5
$C_2H_4$	3.5

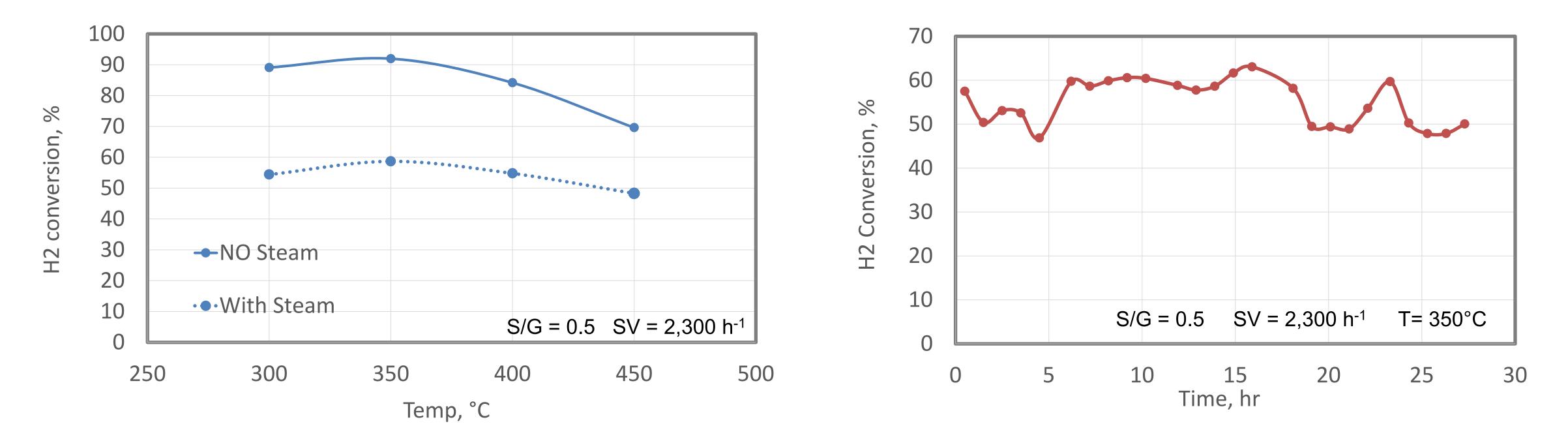
Lab test gas composition

### Methanation

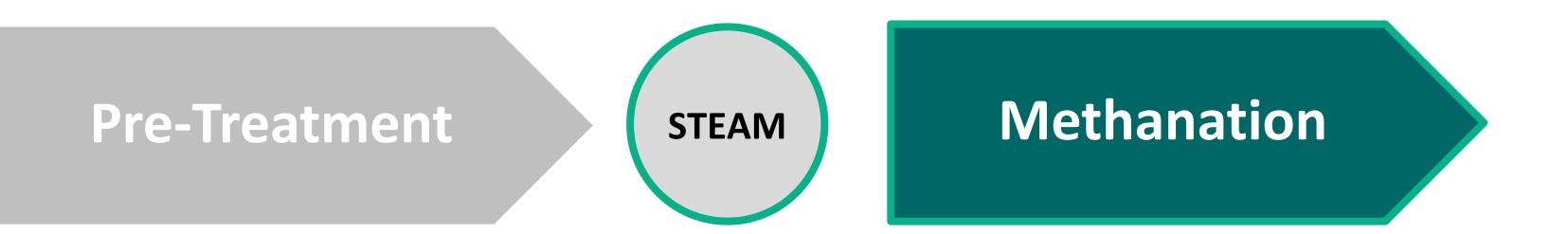




### Steam Reduces Coke But...

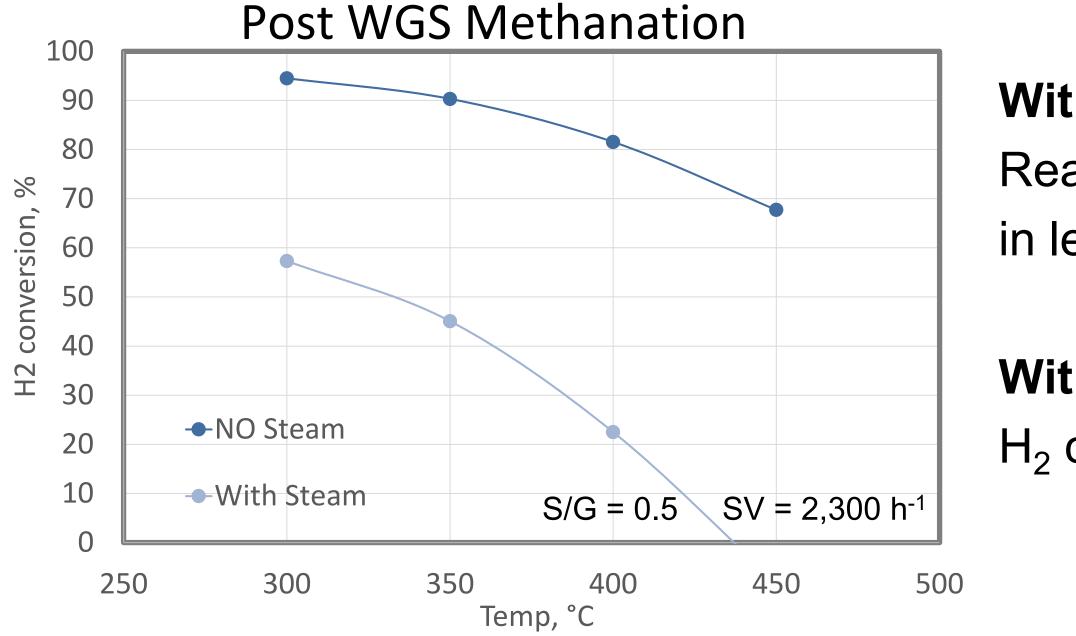


#### \*Coke formations block Reactor after 30hrs





### WGS Yields Greater H<sub>2</sub> & CH<sub>4</sub>







### Without Steam

Reactor blocked with coke in less than 60 hrs

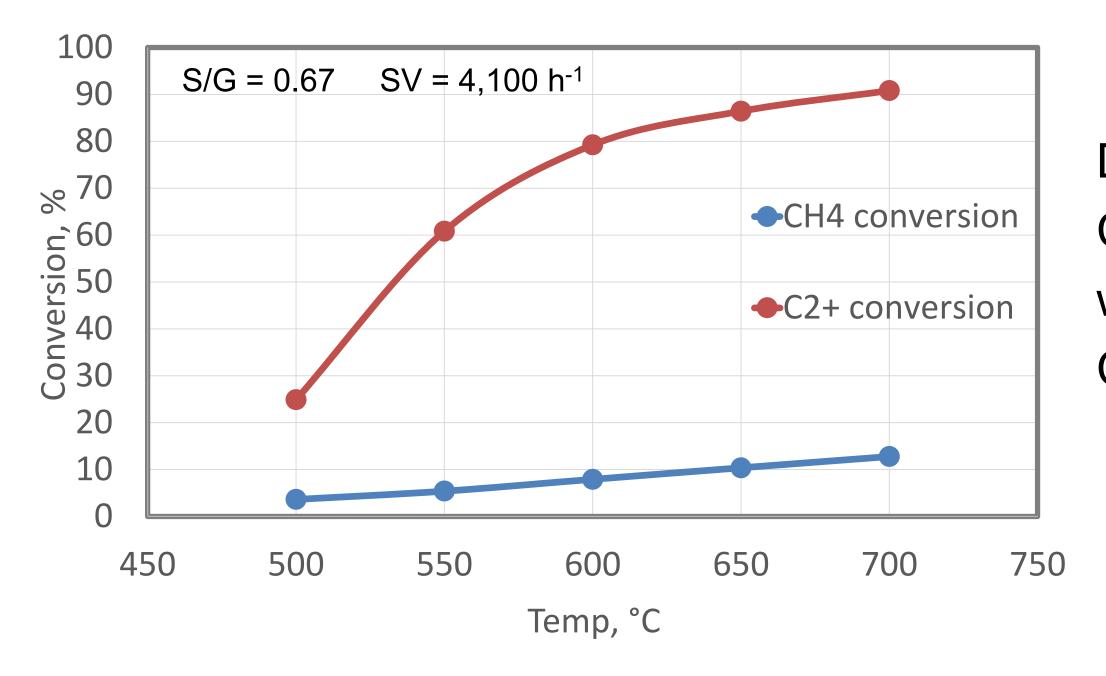
#### With Steam

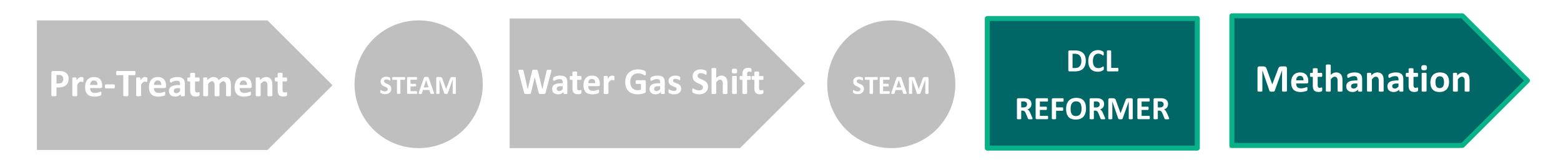
H<sub>2</sub> conversion is very low

Pre- WGS Vol, %	Post- WGS Vol, %
19	40
35	5
21	38
20	13.5
1.5	0.5
3.5	2.5
0	0.5
	WGS   Vol, %   19   355   211   200   1.5   3.5



### Highly Selective Reforming Catalyst



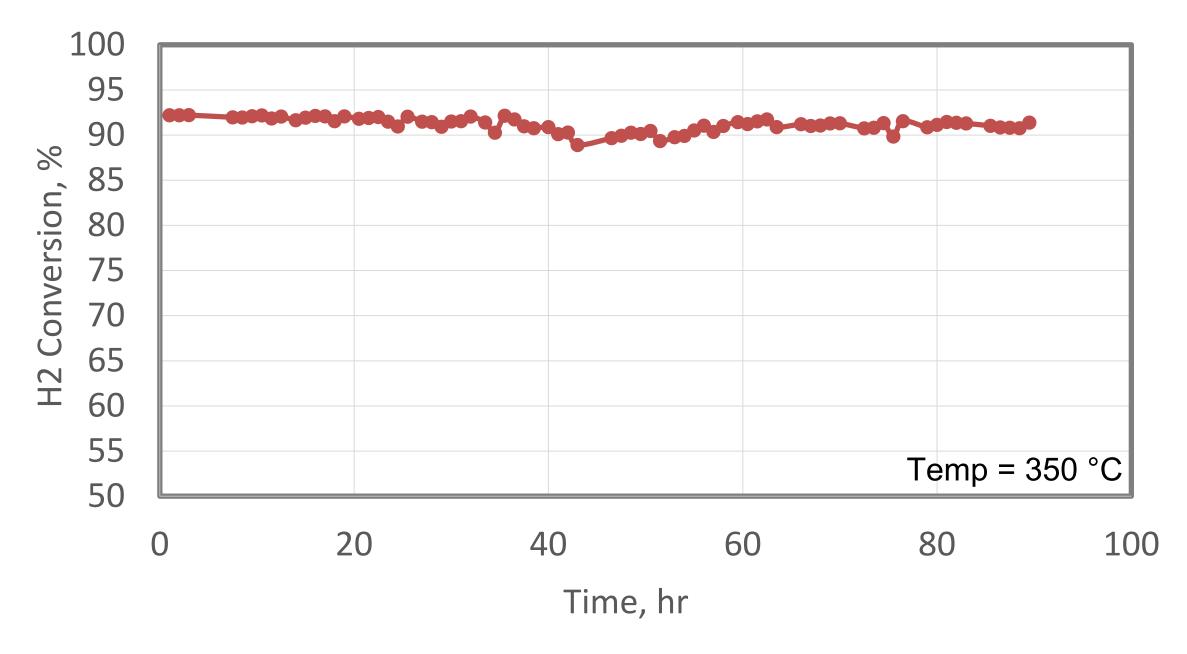


DCL's Proprietary Catalyst can reform  $C_2^+$ without much affecting CH<sub>4</sub> concentration.

Gas Comp.	Post WGS Vol, %	Post Reformer Vol, %
H <sub>2</sub>	40	47
CO	5	11
CO <sub>2</sub>	38	31
$CH_4$	13.5	10
C <sub>2</sub> H <sub>6</sub>	0.5	0.25
$C_2H_4$	2.5	0.002
C <sub>3</sub> H <sub>6</sub>	0.5	0.001



### Methanation of Reformed Gas



Water Gas Shift **Pre-Treatment** STEAM



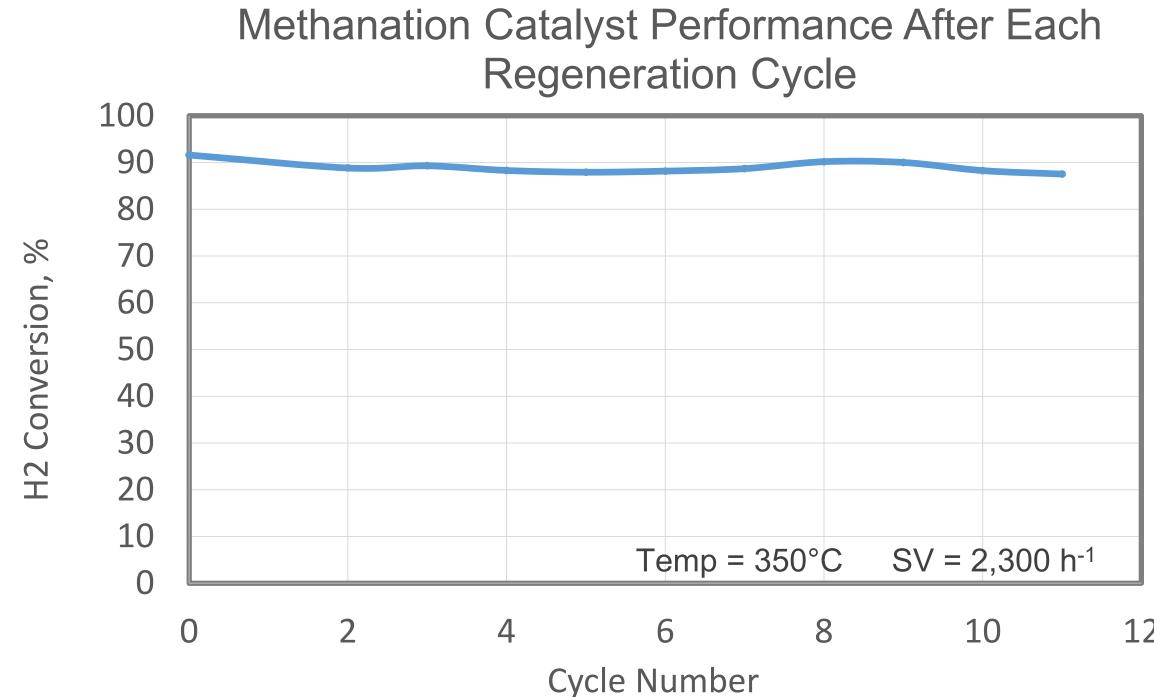
Process	Operating time prior to significant coking	Gas Comp.	Post- Reform Vol, %
Direct	4 hours	H <sub>2</sub>	45.5
Methanation		СО	5
WGS +	<60 hours	CO <sub>2</sub>	44
Methanation		CH <sub>4</sub>	5
WGS + DCL	>100 hours	$C_2H_6$	0.4
<b>Reformer +</b>		$C_2H_4$	0.05
Methanation		C <sub>3</sub> H <sub>6</sub>	0.05

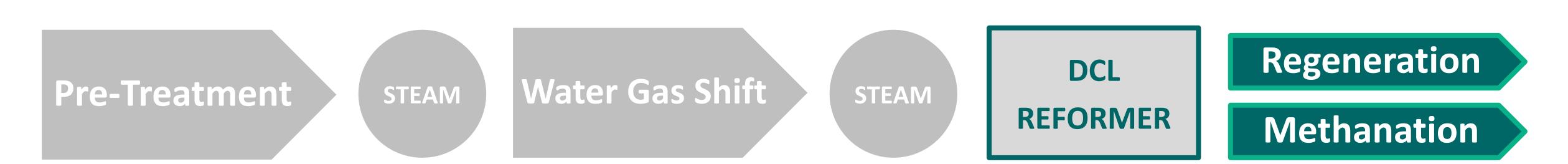
#### \*Coke formations observed after 100 hours of operation.





### Fewer Regeneration Cycles



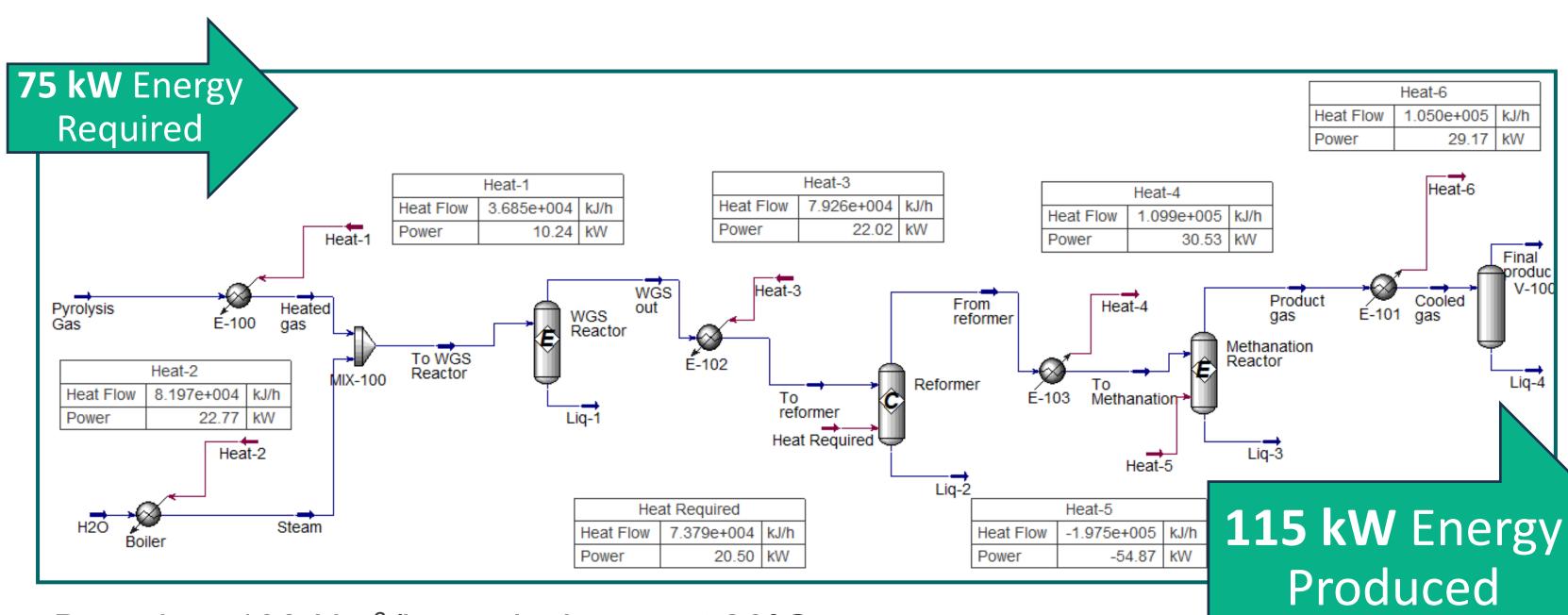




DCL Methanation Catalysts can be regenerated every 4-5 days by running reactors in parallel to effectively burn off coke.

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Gas Comp.	Inlet pyrolysis gas Vol, %	Final product gas Vol, %
H <sub>2</sub>	19	< 5%
CO	35	< 0.5%
CO <sub>2</sub>	21	~50%
$CH_4$	20	~42%
$C_2H_6$	1.5	< 1.0
$C_2H_4$	2.5	< 0.01
$C_3H_6$	1.0	<0.001

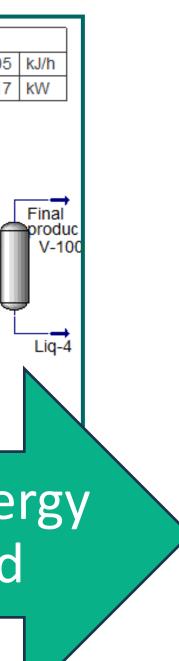


#### Based on 100 Nm<sup>3</sup>/h pyrolysis gas at 20°C

## Final Product & System Design









- DCL's proprietary reforming catalyst enable >90% C2+ hydrocarbon reforming with minimal impact on methane levels.
- DCL reforming catalyst is coke and sulfur resistant.
- DCL reforming catalyst reduces the coke regeneration frequency for the methanation catalyst.







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### Thank You